

# The World of Diverse Matter — The Distant Journey from Space to Earth

The Diversity of Materials Borne from the Collective  
Actions of Atoms, Electrons, and Molecules

## 6<sup>th</sup> Lecture

Quantum World  
Nanoscience, Superconductivity, Superfluidity



Yasuhiro Iye

The Institute for Solid State Physics,  
The University of Tokyo



# Today's Talk

- About Quantum Mechanics
  - Quantum Interference, Tunnel Effect
- Nano-science
  - Mesoscopic Physics
  - Scanning Probe Microscope
- Macroscopic Quantum Phenomena
  - Superfluidity
  - Bose Condensation
  - Superconductivity
- Summary

# Brief Introduction to Quantum Mechanics

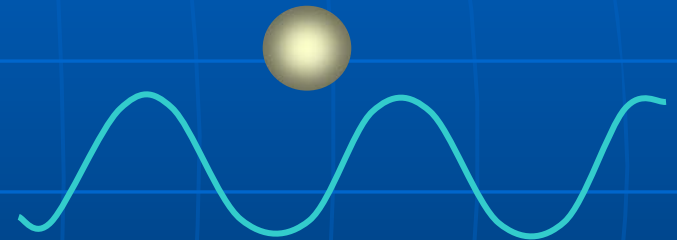
# Quantum Mechanics

Theoretical framework to describe the workings of the micro world

Structure of atoms/molecules  
Behavior of electrons in solids  
Light and Substances

Light: both wave-like and particle-like

Electron: both particle-like and wave-like



The **particle/wave dichotomy** only reflects the fact that we have no suitable expressions when we try to describe quantum mechanical behaviors in terms of everyday (classical mechanical) words.

Particle nature: discrete, countable

Wave nature: superposition, interference

# Quantum Mechanical Particles also Exhibits Wave-like Behavior

de Broglie Wavelength

$$\lambda = \frac{h}{p}$$

Momentum

The de Broglie wavelength of an electron that is accelerated at 100V

$$E = \frac{p^2}{2m} \Rightarrow p = \sqrt{2mE}$$

$$\begin{aligned}\lambda &= \frac{h}{p} = \frac{h}{\sqrt{2mE}} \\ &= \frac{6.62 \times 10^{-34}}{\sqrt{2 \times 0.91 \times 10^{-30} \times 100 \times 1.6 \times 10^{-19}}} \\ &= 1.23 \times 10^{-10} \text{ m} \quad \lambda = 0.12 \text{ nm}\end{aligned}$$

The de Broglie wavelength of a classical particle (e.g. a tennis ball) is *extreeeemely* short

# Wave Function

The state of a quantum mechanical particle is described by a wave function

$$\psi(x, y, z, t)$$

The probability of finding the particle at a position  $(x, y, z)$  at time  $t$  is given by

$$|\psi(x, y, z, t)|^2$$

The time evolution of a wave function obeys Schrödinger's equation

$$i\hbar \frac{d}{dt} \psi(x, y, z, t) = \left( -\frac{\hbar^2}{2m} \left( \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right) + V(r) \right) \psi(x, y, z, t)$$

This is a linear equation => Superposition Principle

$$\psi_1, \psi_2 \Rightarrow \psi_1 + \psi_2$$

=> Quantum Interference

# The Role of Measurements in Quantum Mechanics

Quantum mechanics provides a probabilistic distribution of the outcome of the measurements of a physical quality when they are repeated in the "same situation". However, it does not generally give a definite value for an individual measurement.

Immediately after a measurement, the state takes one of the eigenstates of the physical observable.

=> "Collapse of state"

The unitary time evolution is described by Schrödinger's equation, whereas measurements give rise to collapse of the state.

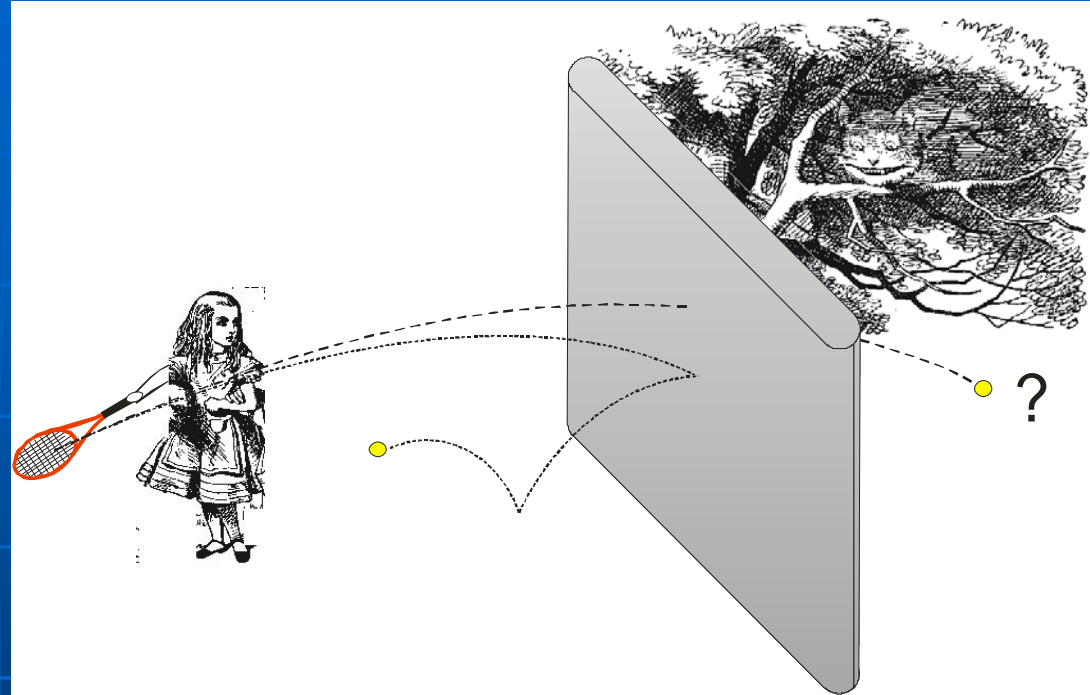
The standard interpretation of quantum mechanics  
(Copenhagen Interpretation)

Interpretation problem, observation problem of quantum mechanics  
Many-Worlds Interpretation

# Characteristic Quantum Mechanical Phenomena

## Tunnel Effect

- Quantum mechanical particles can go through a potential wall they cannot penetrate according to classical dynamics

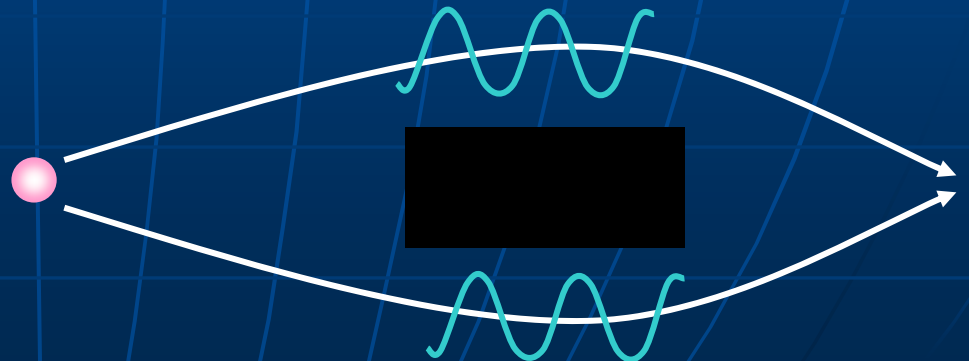


Yasuhiro Iye 2006 "Alice's quantum mechanics"(In Japanese) Maruzen†

## Quantum Interference Effect

- Superposition of states that pass through different paths

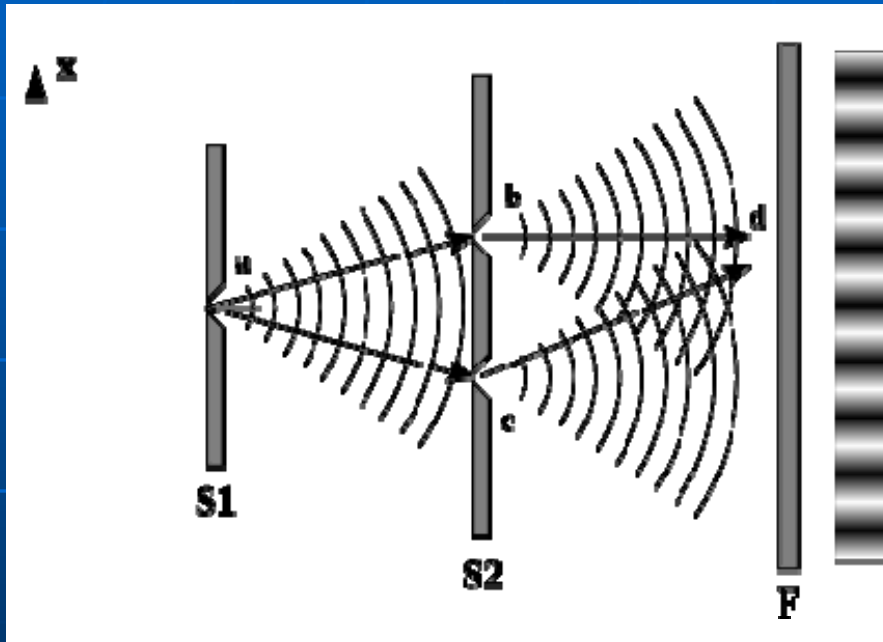
=> Quantum Interference



# Quantum Interference

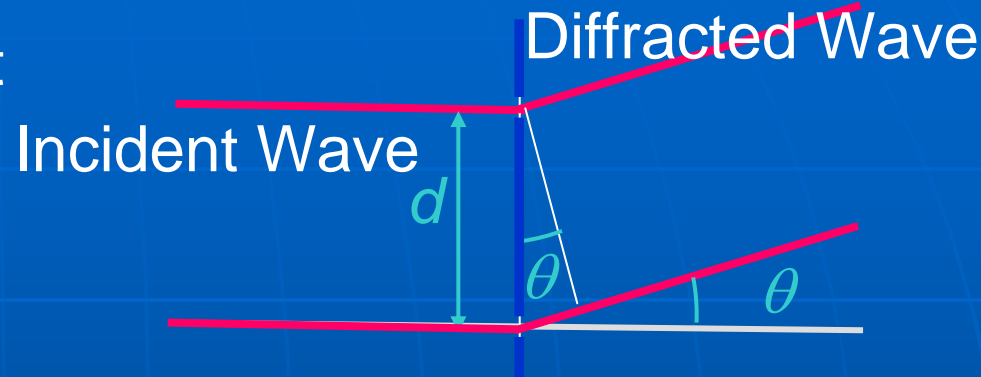
# Interference of Light Waves

Young's double-slit experiment  
(1805)

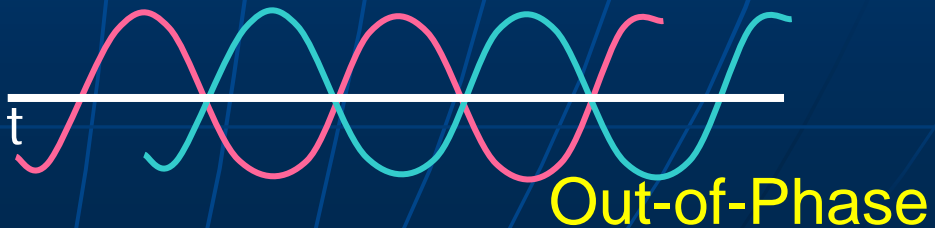
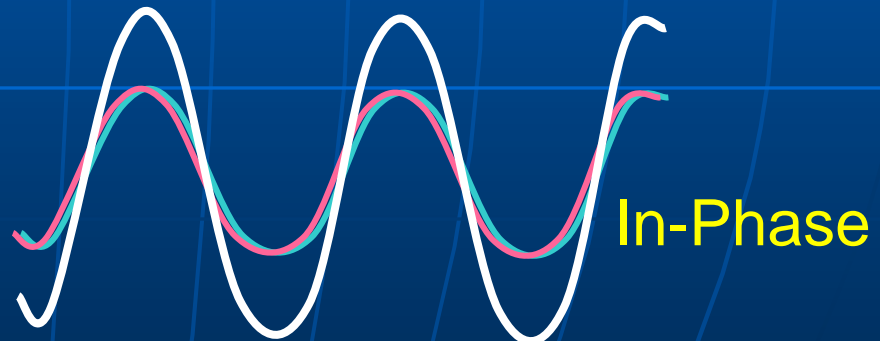


<http://en.wikipedia.org/wiki/File:Ebohr1.svg>

Interference of waves  
that passed through a double slit

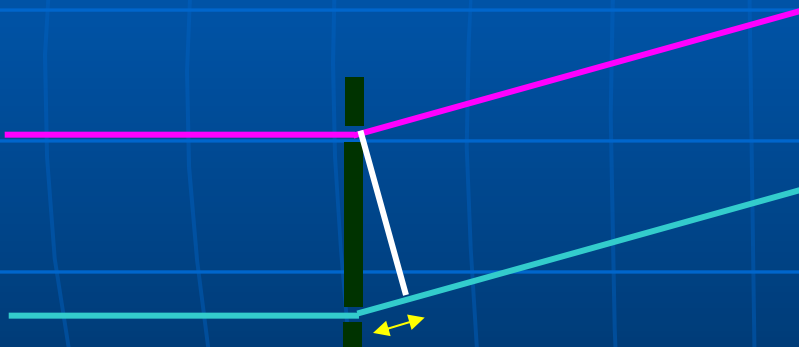


$$d \sin \theta = \begin{cases} n\lambda & \text{enhance} \\ (n + \frac{1}{2})\lambda & \text{cancel} \end{cases}$$

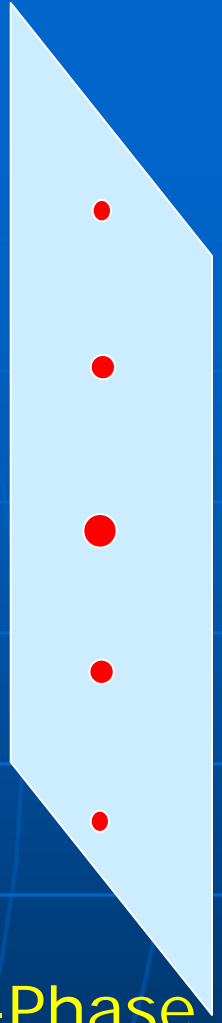
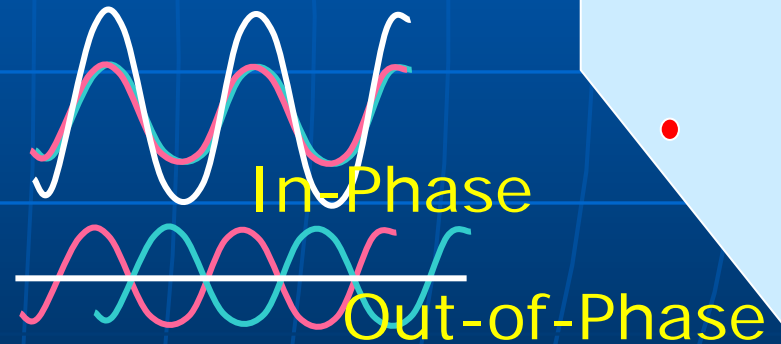


# Diffraction of Light

## Diffraction



Optical Path Difference

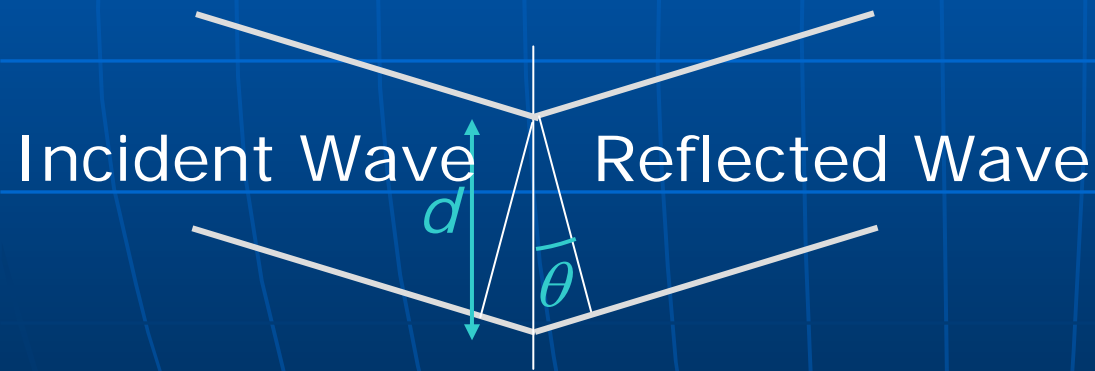
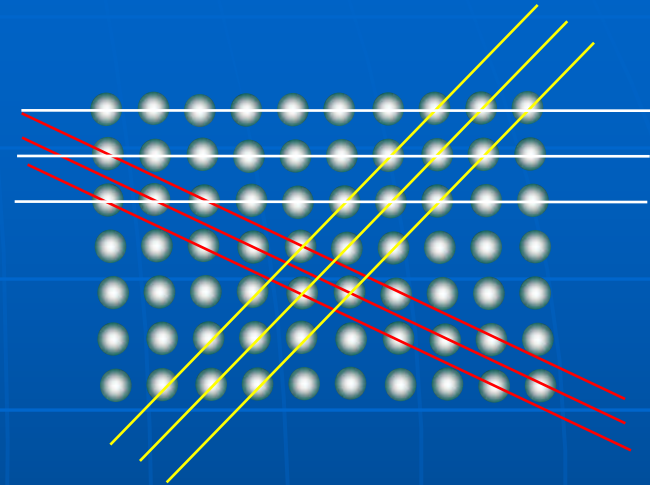
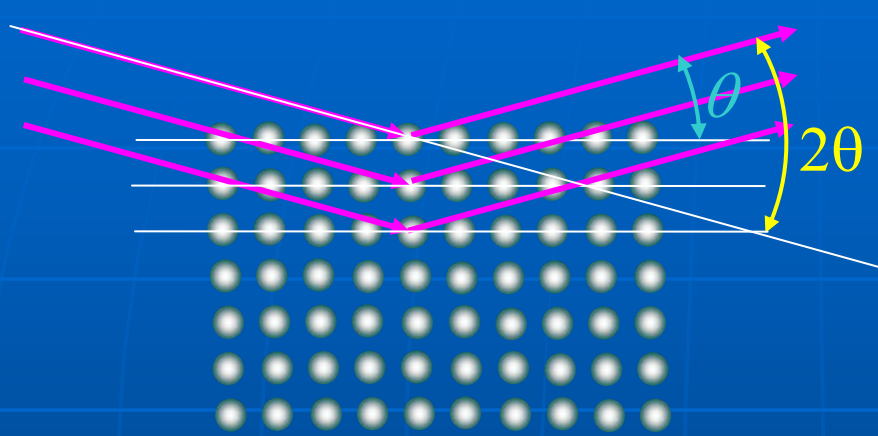


$$d \sin \theta = n\lambda$$

Inter-atomic Spacing  $\sim 0.3\text{nm}$   $\sim$  Wavelength of X-Rays

Diffraction experiment can be also done with electron beams and neutron beams.

# Crystal Structure Analysis



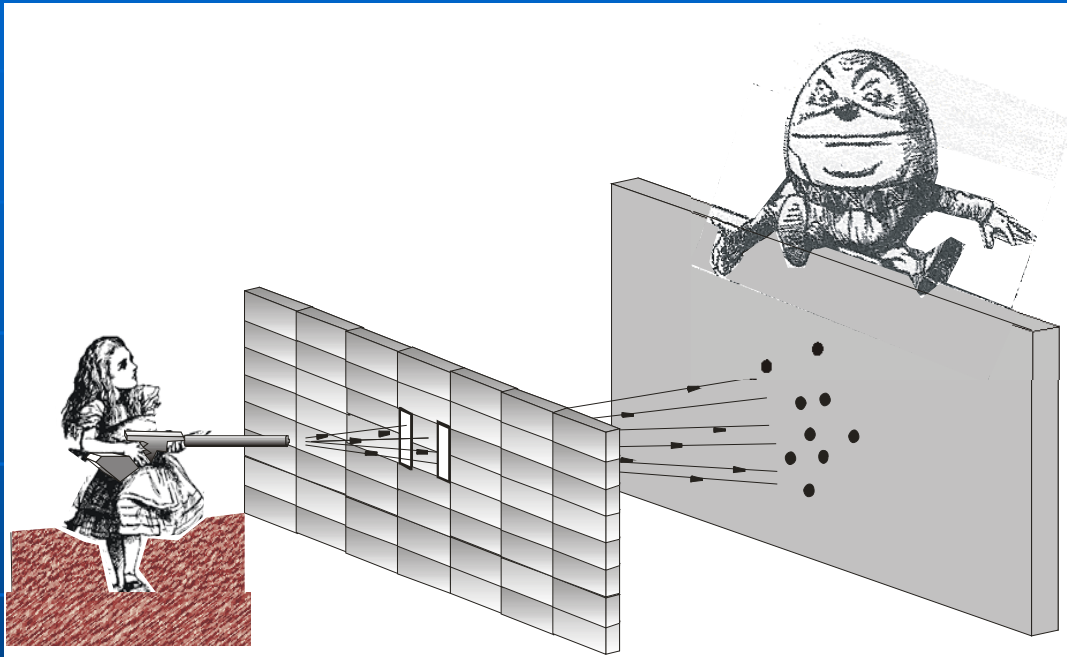
$$2d \sin \theta = n\lambda$$

Bragg Condition



Four-Axes X-ray Diffractometer

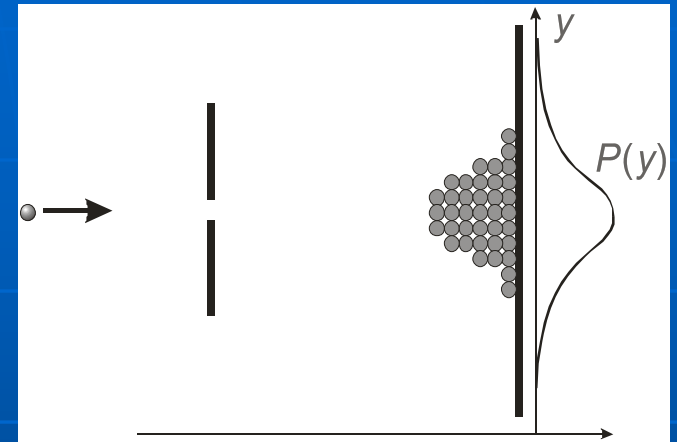
# The Case of Classical Particles



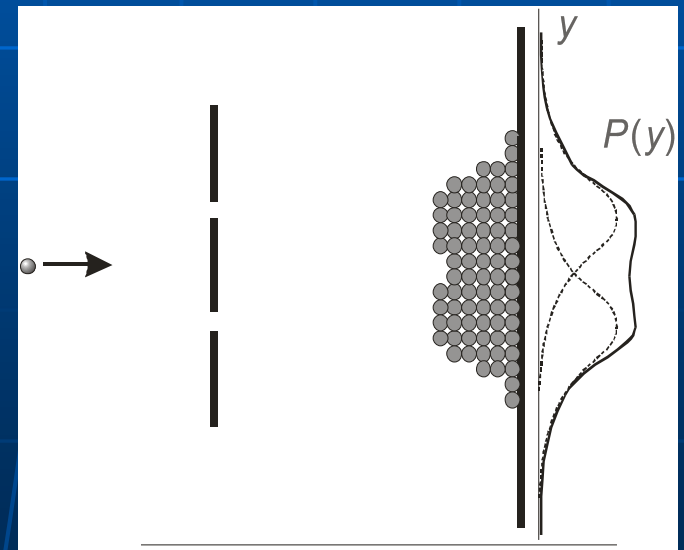
Yasuhiro Iye 2006 "Alice's quantum mechanics"(In Japanese) Maruzen

Probability of a bullet hitting a particular spot  
= Probability of reaching the spot via the right slit  
+ Probability of reaching the spot via the left slit

1 slit



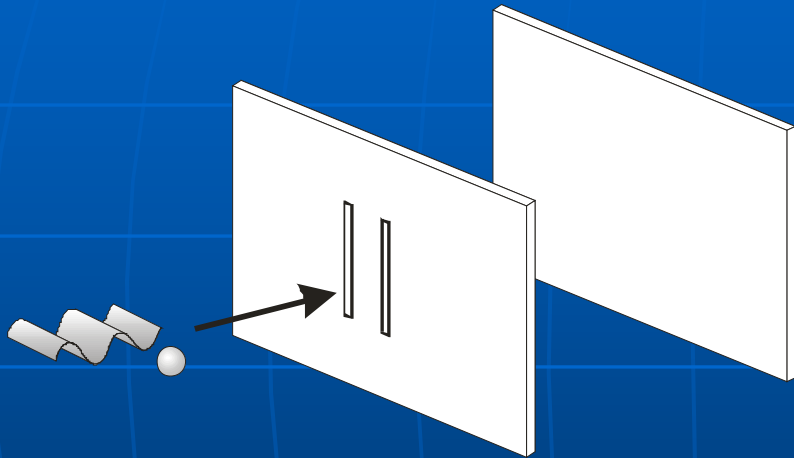
2 slits



$$P_{\text{total}}(y) = P_{\text{R}}(y) + P_{\text{L}}(y)$$

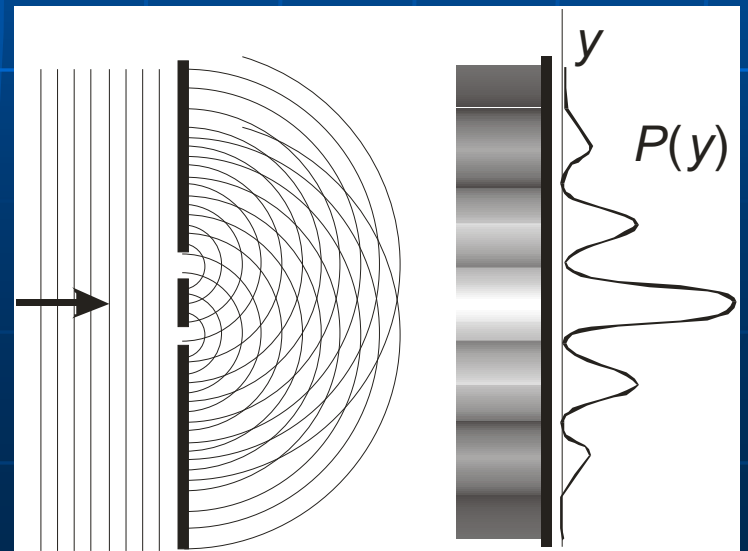
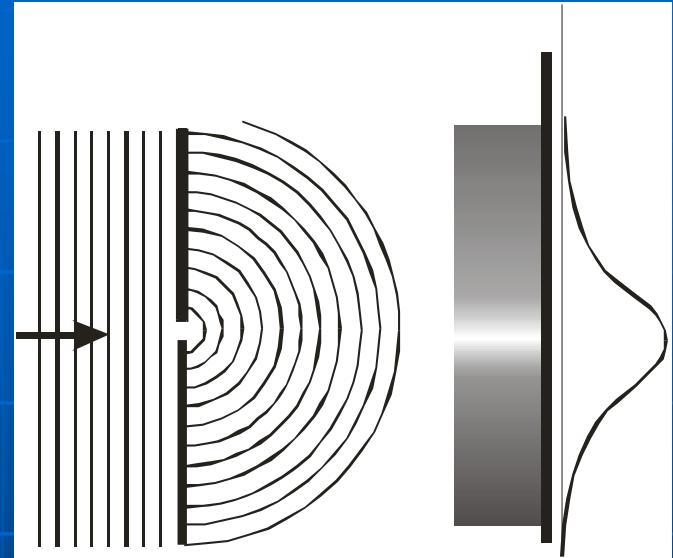
Summation of Probability

# The Case of Waves



Summation of  
Wave Amplitudes

Interference Pattern

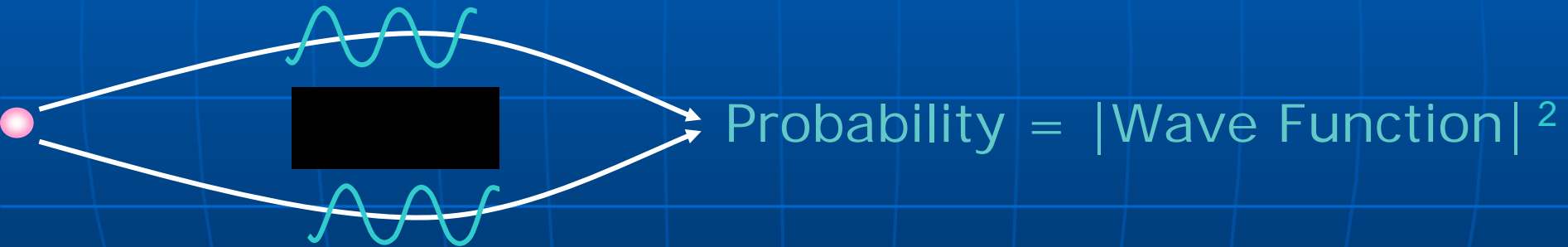


# The Case of Quantum Mechanical Particles

$$\Psi_{\text{total}} = \Psi_{\text{R}} + \Psi_{\text{L}}$$

Total Wave Function

= Wave Function passing through right-hand slit  
+ Wave Function passing through left-hand slit



$$\begin{aligned} |\Psi_{\text{total}}|^2 &= |\Psi_{\text{R}} + \Psi_{\text{L}}|^2 \\ &= |\Psi_{\text{R}}|^2 + |\Psi_{\text{L}}|^2 + \underbrace{\Psi_{\text{R}}^* \Psi_{\text{L}} + \Psi_{\text{R}} \Psi_{\text{L}}^*}_{\text{Quantum Interference Terms}} \end{aligned}$$

**Quantum Interference Terms**

# Interference of Electrons

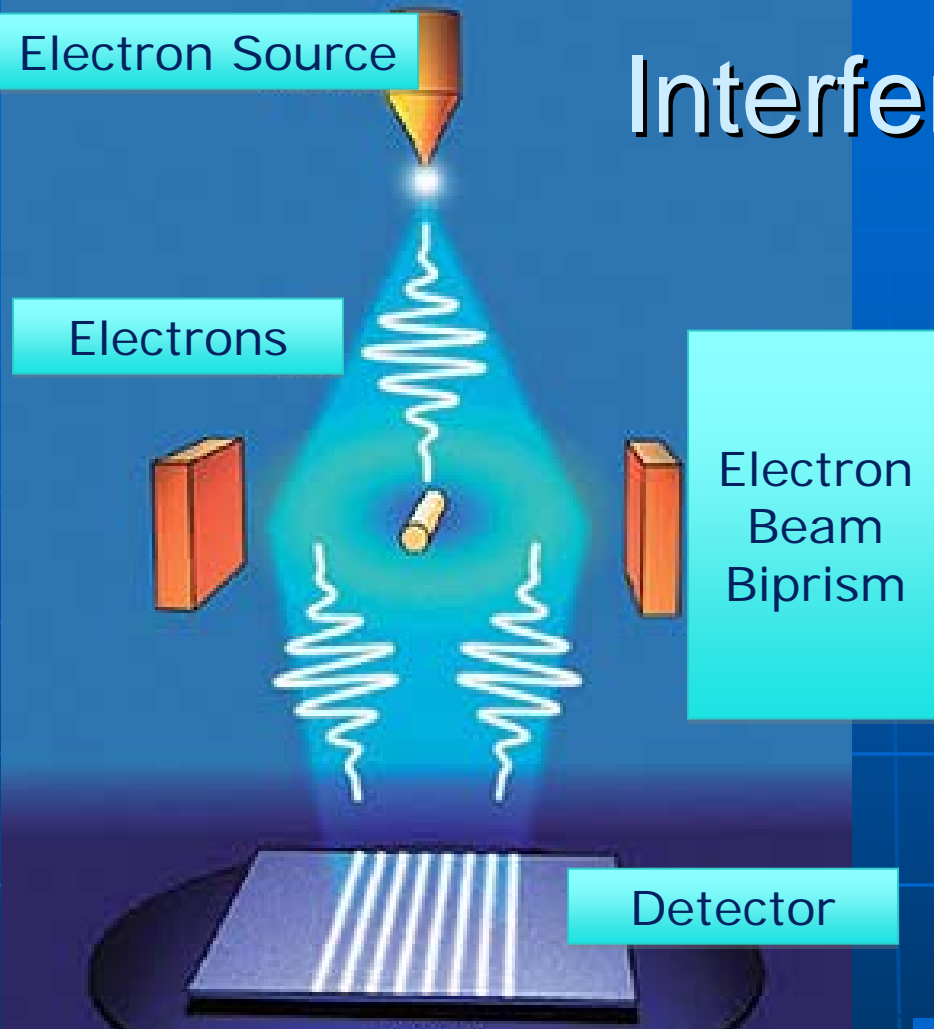
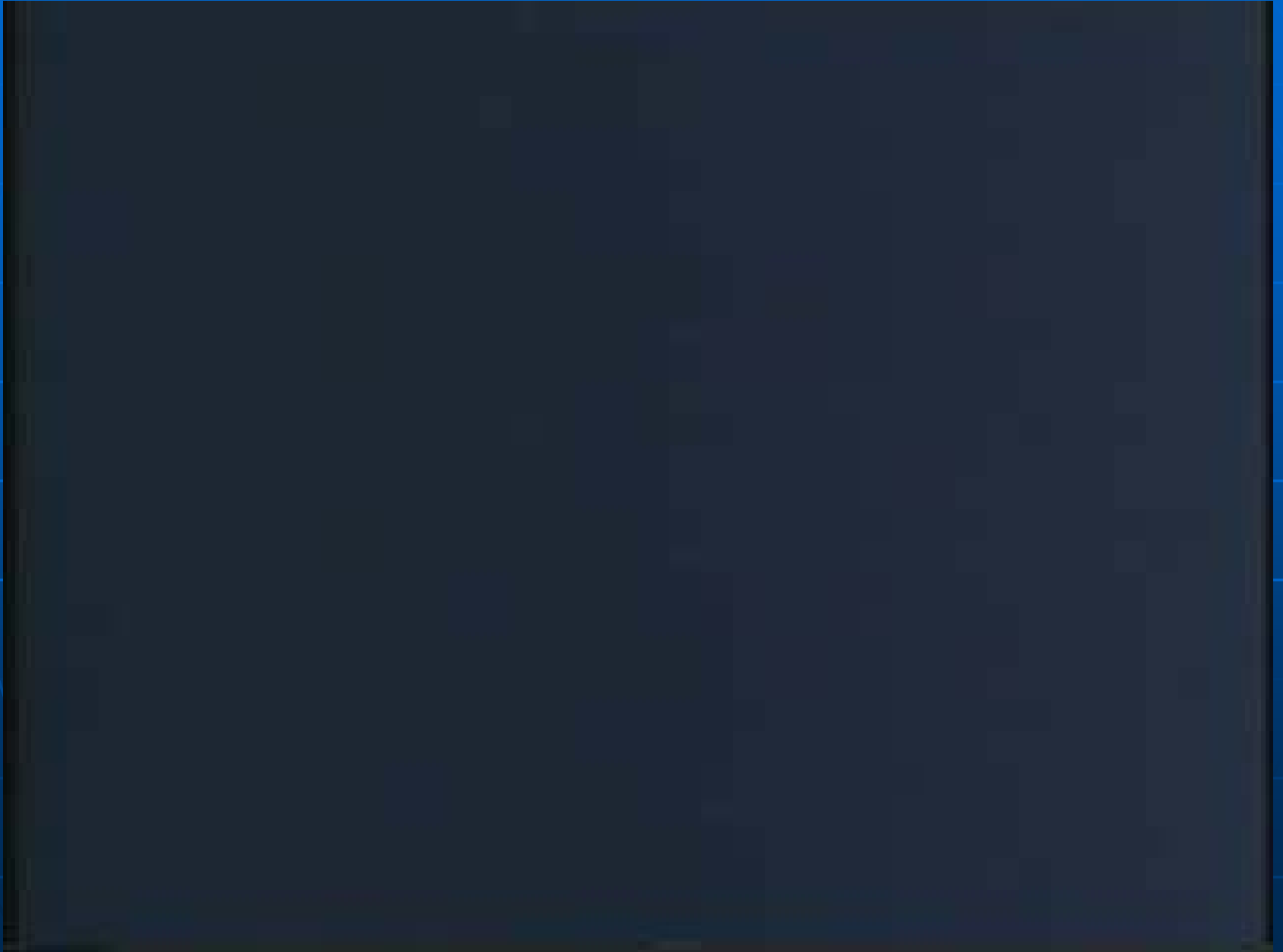


Fig. 3 Representation of Electron Two-Slit Experiment

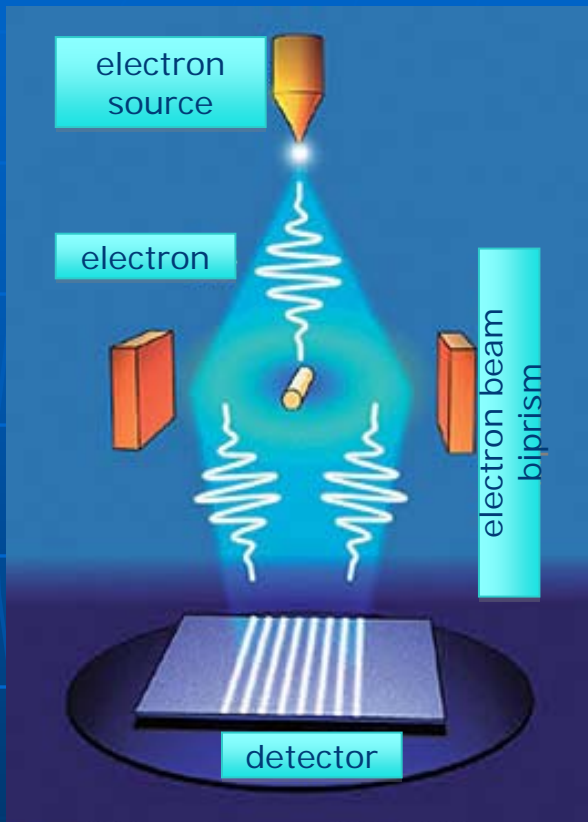
In this experiment an electron beam bi-prism is used in place of a slit. This type of experiment is technically difficult and at one time was thought of as a thought experiment that could only be done in your head.

Dr Akira Tonomura  
(Hitachi Advanced  
Research Laboratory) †

# Double-Slit Experiment with Electrons (Dr Akira Tonomura)



# Interference of Electrons

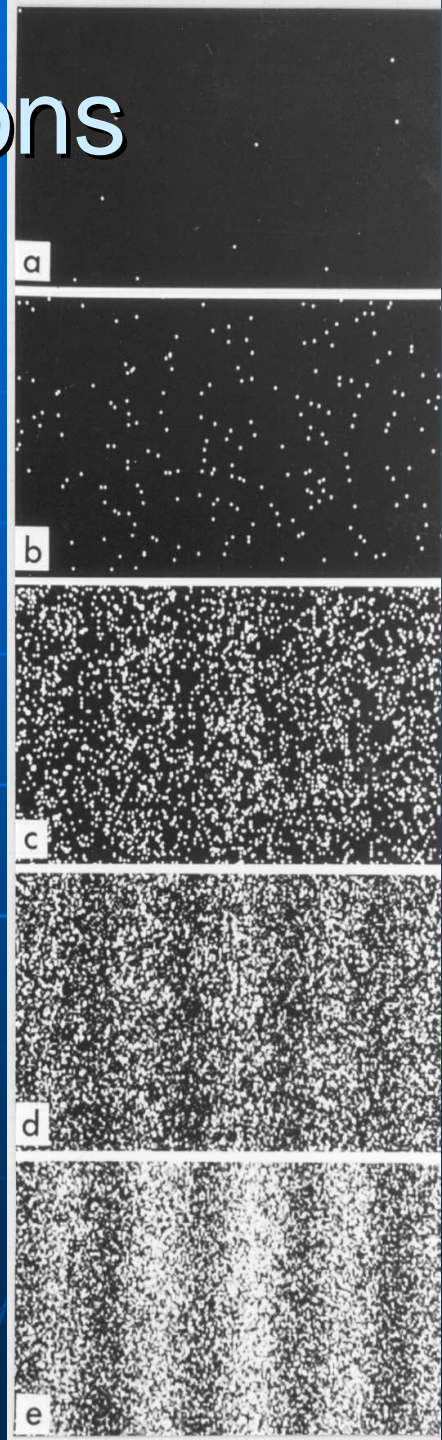


Dr Akira Tonomura  
(Hitachi Advanced  
Research Laboratory)

Electrons reach  
the screen one-  
by-one

Interference pattern  
emerges

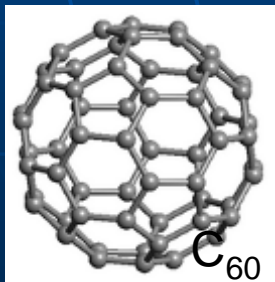
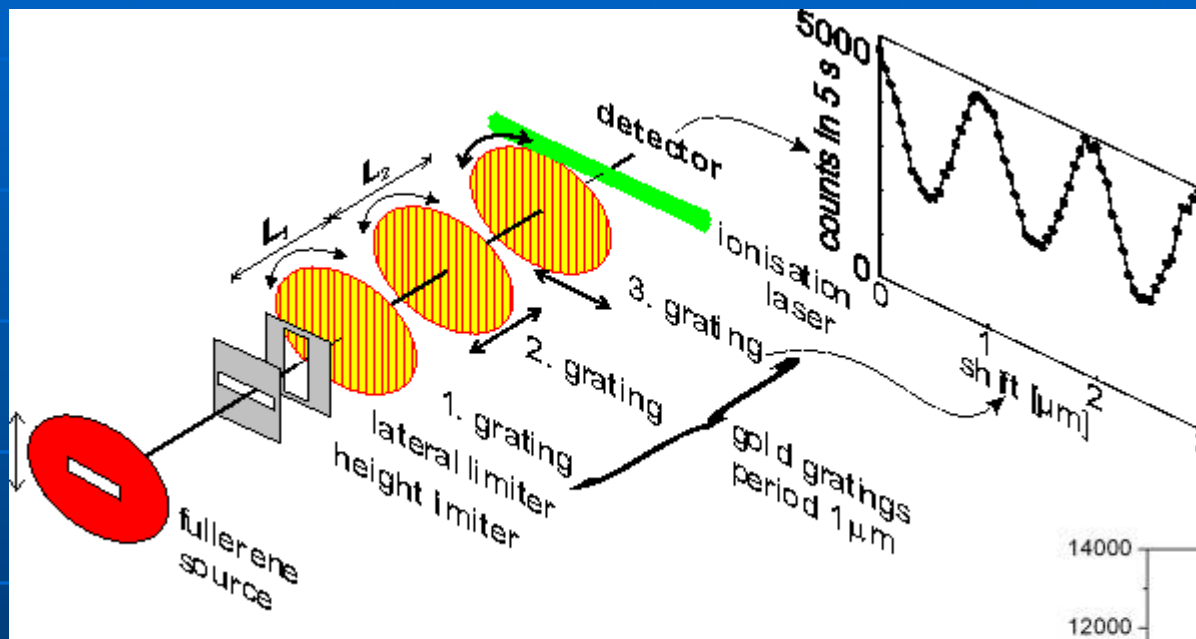
Unambiguous proof of  
the wave nature of  
electron.



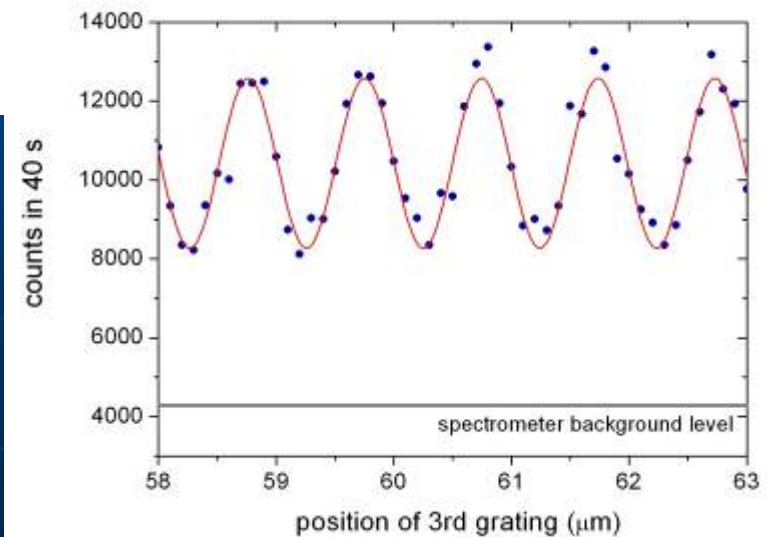
# How Large an Object Can Still Interfere?

A. Zeilinger

Vienna University of Technology



C<sub>60</sub> Fullerene molecule

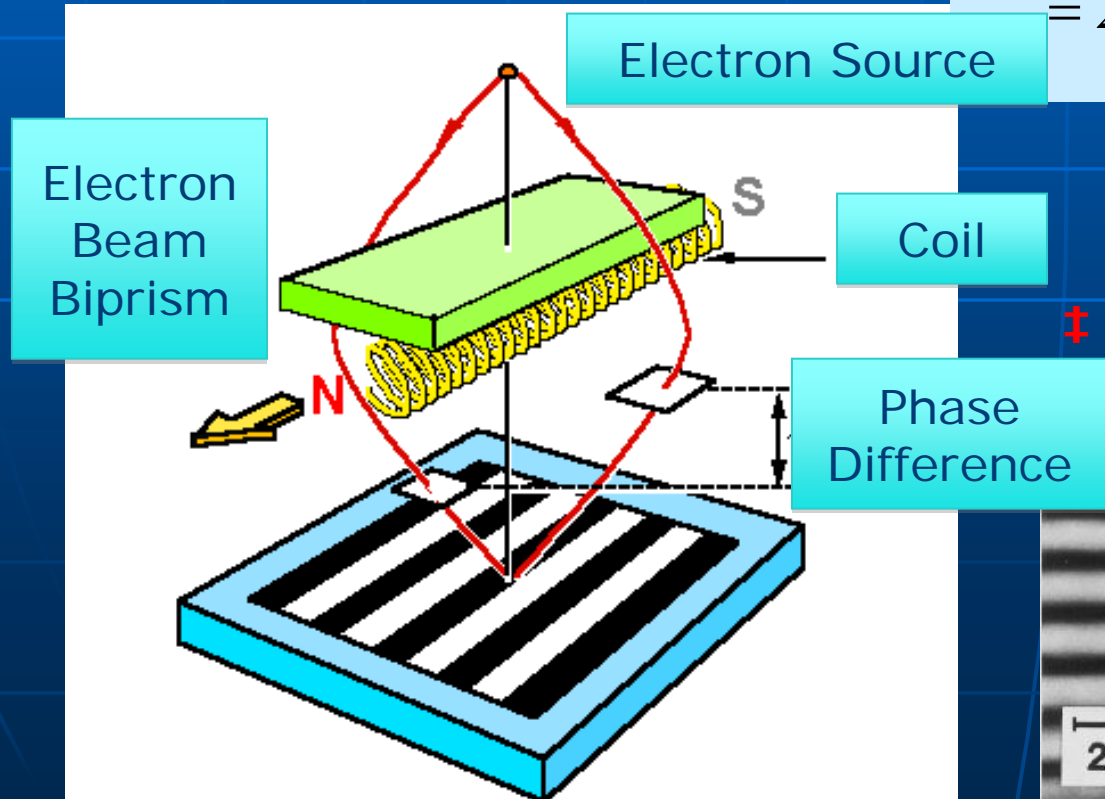


# Aharonov-Bohm (AB) Effect

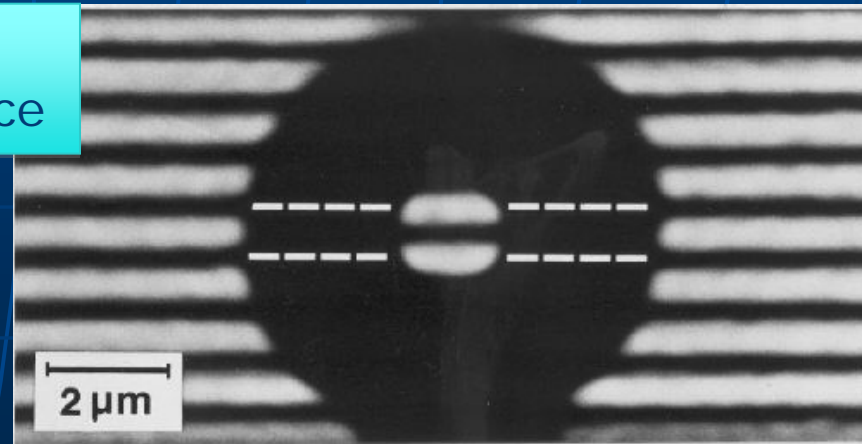
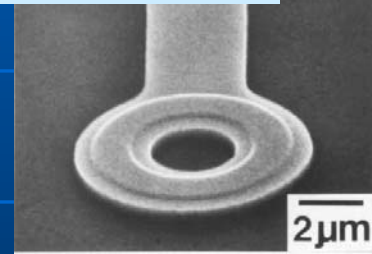
The phase of electron wave changes with magnetic field (more precisely, with vector potential)

$$\psi \Rightarrow \psi e^{i \frac{e}{\hbar} \int \mathbf{A}(\mathbf{r}) \cdot d\mathbf{r}}$$

$$\begin{aligned} \Delta\theta &= \frac{\hbar}{e} \int_L \mathbf{A}(\mathbf{r}) \cdot d\mathbf{r} - \frac{\hbar}{e} \int_R \mathbf{A}(\mathbf{r}) \cdot d\mathbf{r} \\ &= \frac{\hbar}{e} \oint_{\text{loop}} \mathbf{A}(\mathbf{r}) \cdot d\mathbf{r} = \frac{\hbar}{e} \int \mathbf{B}(\mathbf{r}) \cdot d\mathbf{S} \\ &= 2\pi \frac{\phi}{\phi_0} \quad \phi_0 = \frac{h}{e} \end{aligned}$$



Dr Akira Tonomura  
(Hitachi Advanced  
Research Laboratory)



Mesoscopic System

Nanotechnology

# Mesoscopic Physics



Intermediate range between micro and macro  
(The prefix “meso” stands for intermediate)

Physics characteristic to the systems whose scale is comparable to or smaller than the “characteristic length scale of the relevant physical phenomenon”  
=> mesoscopic physics

Examples of the “characteristic” length scale:

Wavelength of an electron (Fermi wavelength)

Mean free path of an electron (average distance an electron travels before it is scattered)

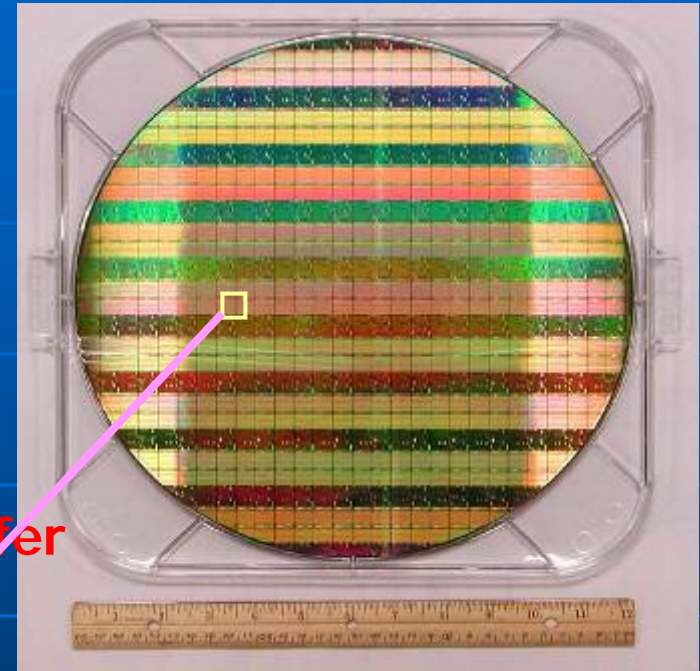
Phase relaxation length of an electron

# Miniaturisation of Semiconductor Devices

<http://en.wikipedia.org/wiki/File:Replica-of-first-transistor.jpg>



50 years



Silicon Wafer

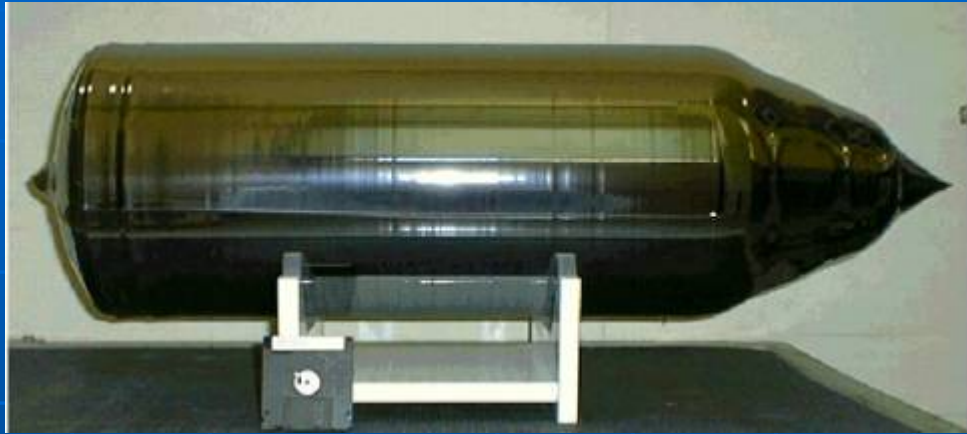
12 inches (30 cm)

First Electrical Transistor  
Bell Telephone Laboratories  
(1946)

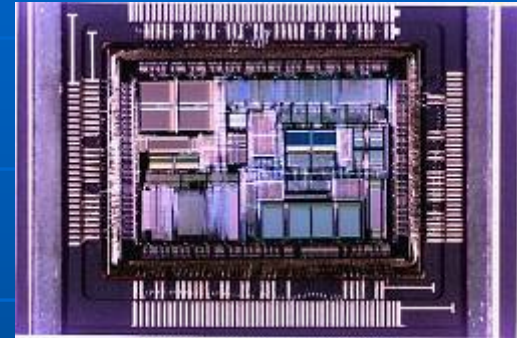


Progress in Semiconductor  
High-Tech Industry

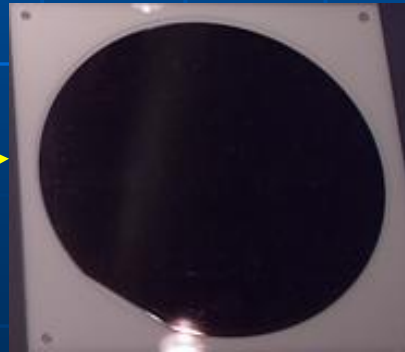
# Silicon Single Crystal => Wafer => ULSI



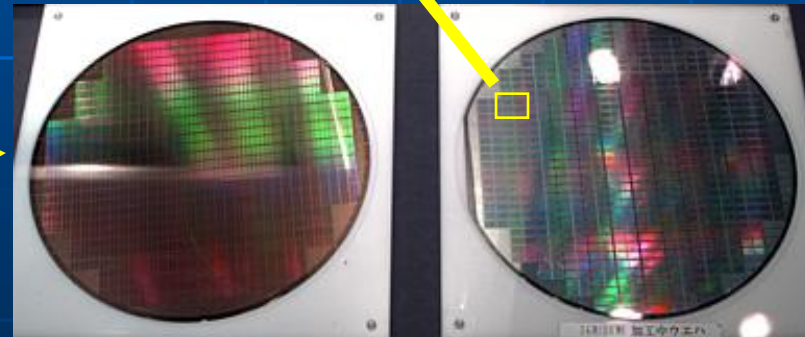
ULSI  
(Ultra Large-Scale Integrated Circuit)



Slicing and Polishing

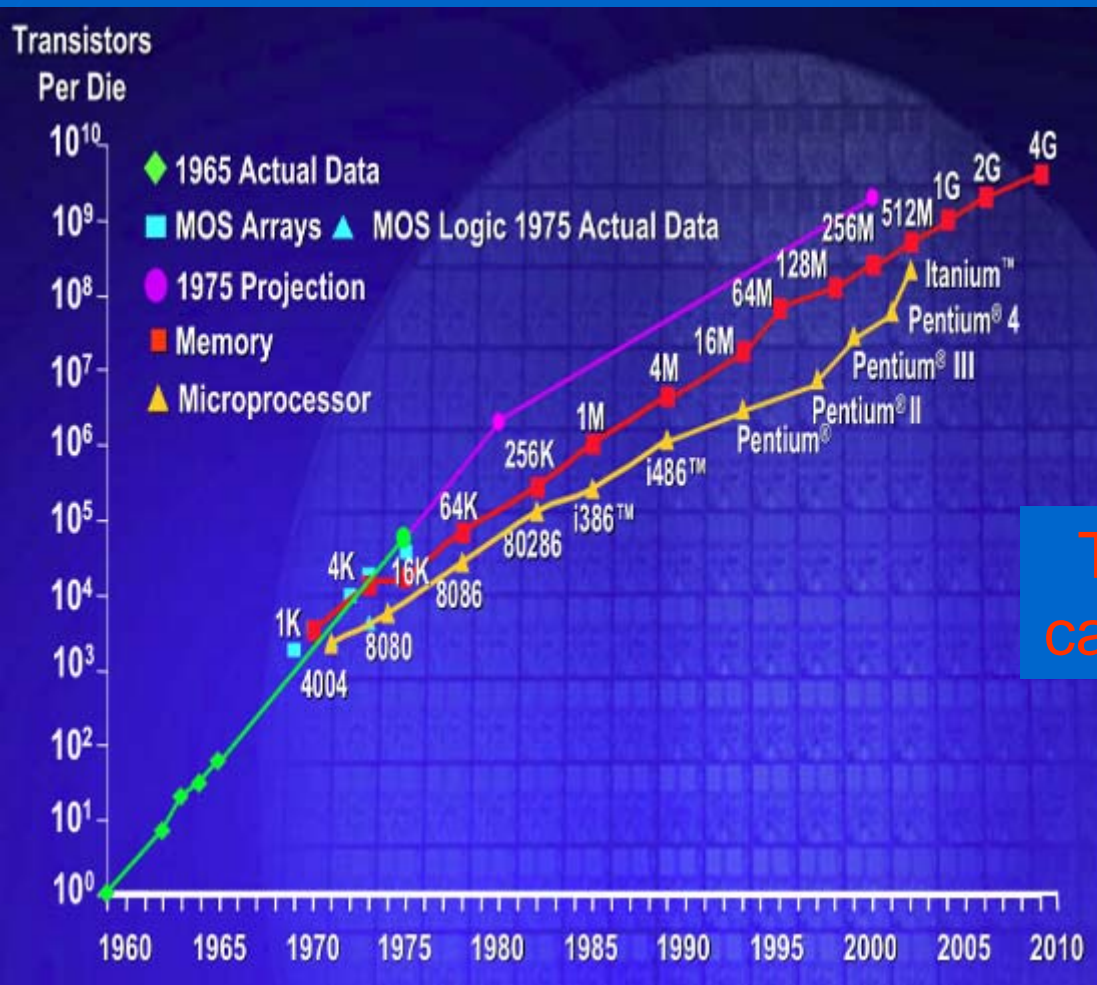


Wafer



Microfabrication

# Moore's Law



Gordon Moore (1965)  
Co-founder of Intel

Typical example of the so-called self-fulfilling prediction

Gordon E. Moore "NO EXPONENTIAL IS FOREVER..."

†Intel Copyright Permission Department

The degree of integration of LSI (Large Scale Integrated Circuit), namely the number of transistors that can be packed into a given area of semiconductor chip doubles every one-and-a-half or two years.

# Nano-science / Nanotechnology

- Miniaturization technology  $\Rightarrow$  **More than Moore**

It is not a simple scaling down

- Problem associated with heat generation
- Problem associated with quantum fluctuation

- Nano-science

- Observing atoms, Manipulating atoms
- Pursuit of new quantum effects in the nano-world

**“There is plenty of room at the bottom.”**

(Richard Feynman)

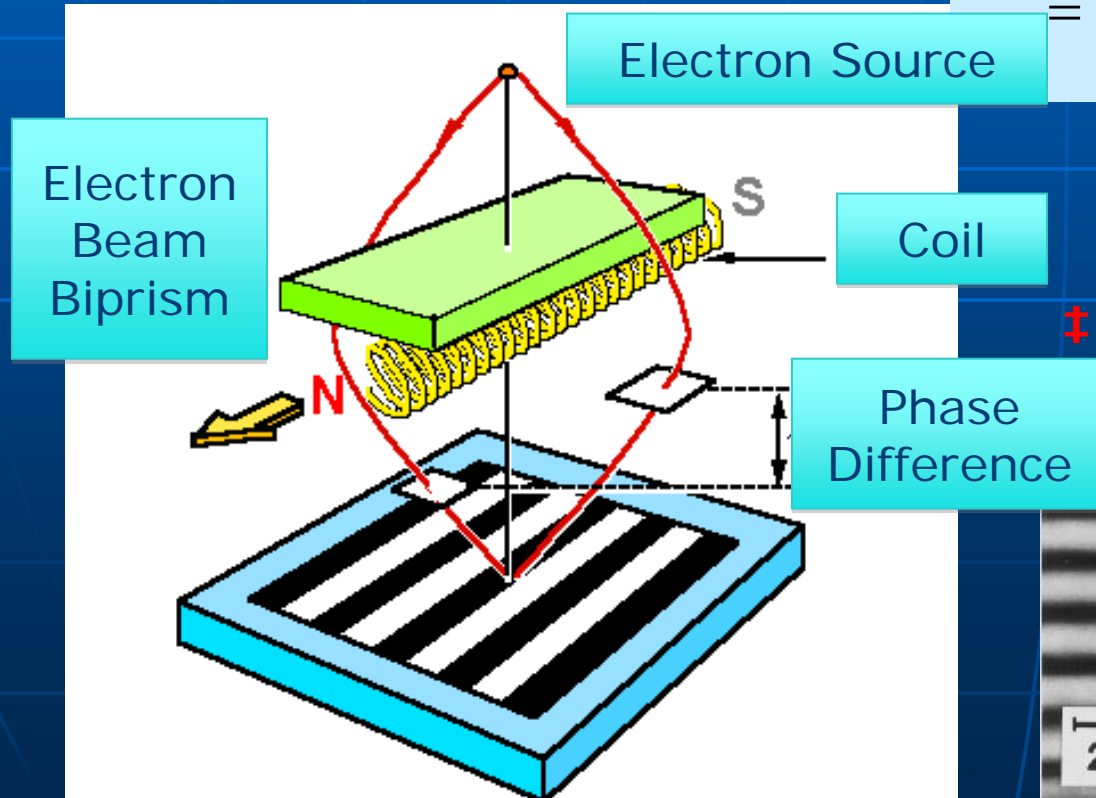
# Mesoscopic Physics

# Aharonov-Bohm (AB) Effect

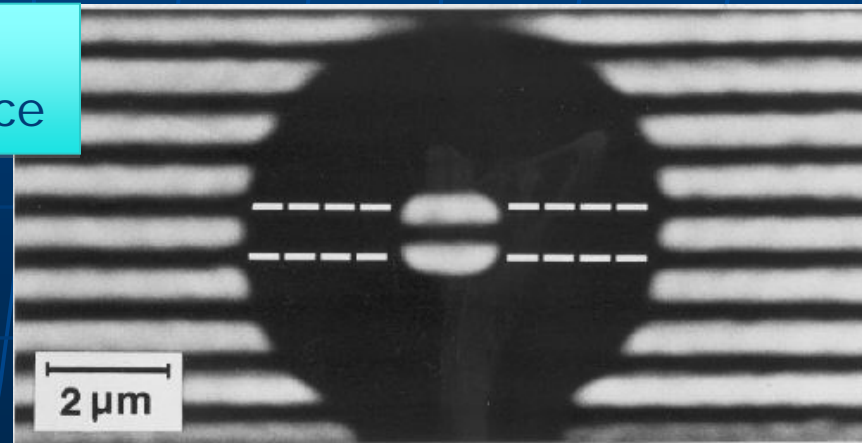
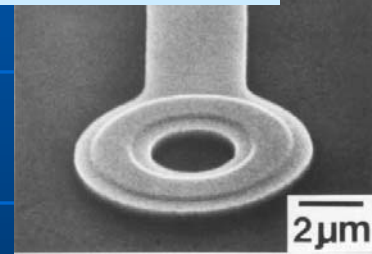
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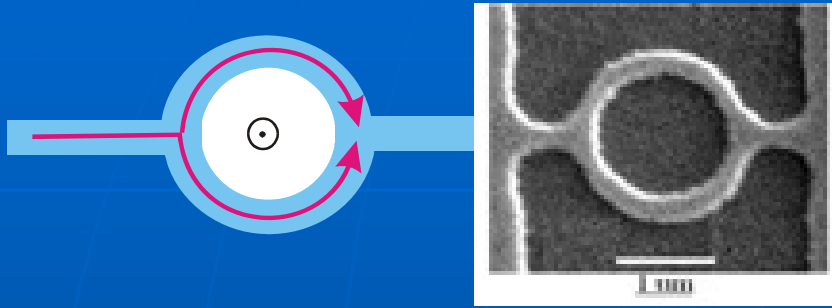
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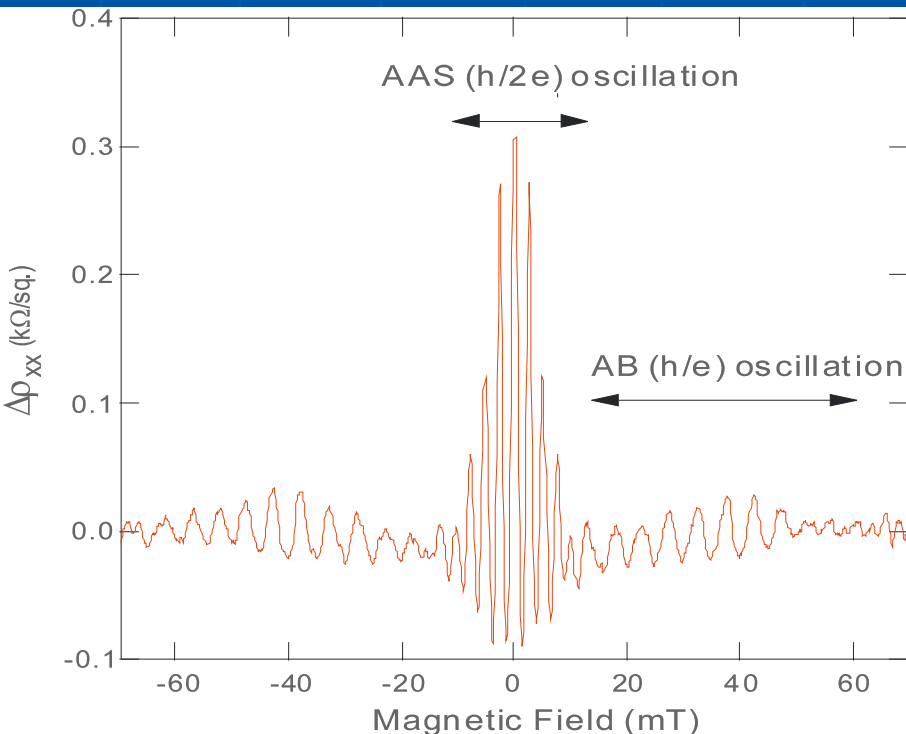
# Aharonov-Bohm Oscillations



Mesoscopic Ring

$$\begin{aligned}
 |\Psi|^2 &= |\Psi_+ + \Psi_-|^2 \\
 &= |\Psi_+|^2 + |\Psi_-|^2 + \Psi_+ \Psi_-^* + \Psi_+^* \Psi_-
 \end{aligned}$$

quantum interference term



Interference between two electron waves that travel along the paths on both sides of a ring.

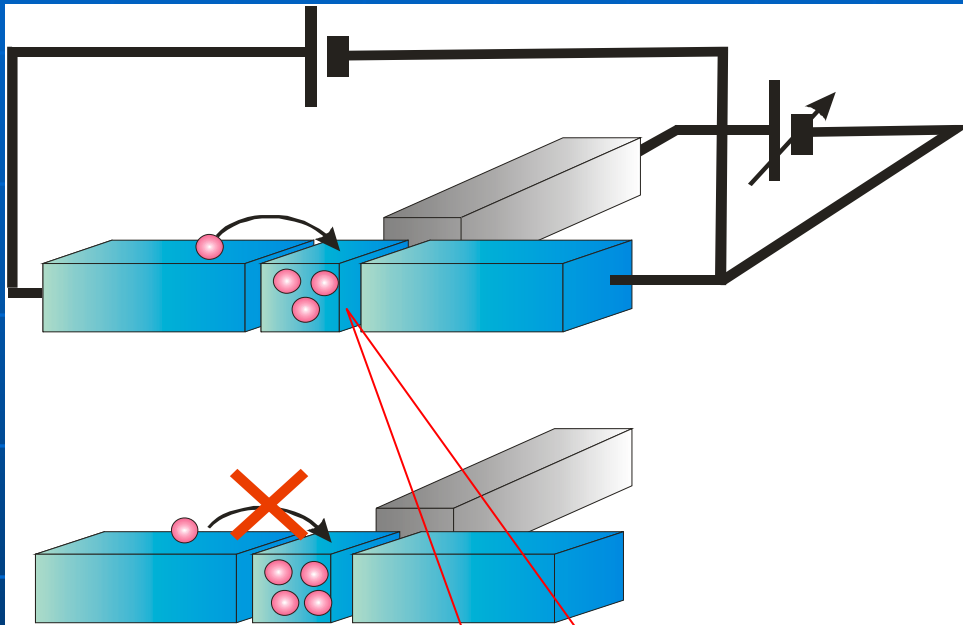
Electrical resistance changes periodically with the magnetic flux  $\Phi$  piercing the ring.

$$\phi_0 = \frac{h}{e} = 4.14 \times 10^{-15} \text{ Wb}$$

Flux Quantum

# Single Electron Tunnelling

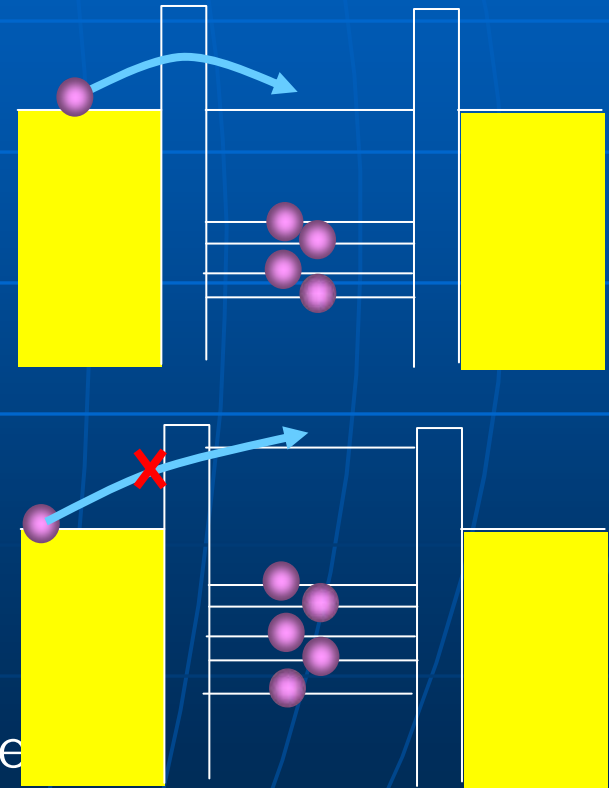
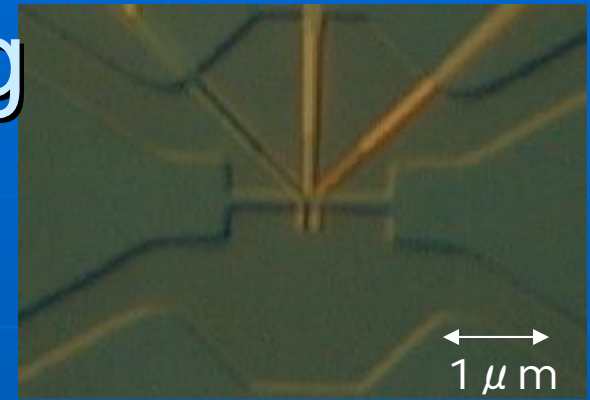
## Nano-Tunnel Junction



Coulomb Island  
(Quantum Dot)

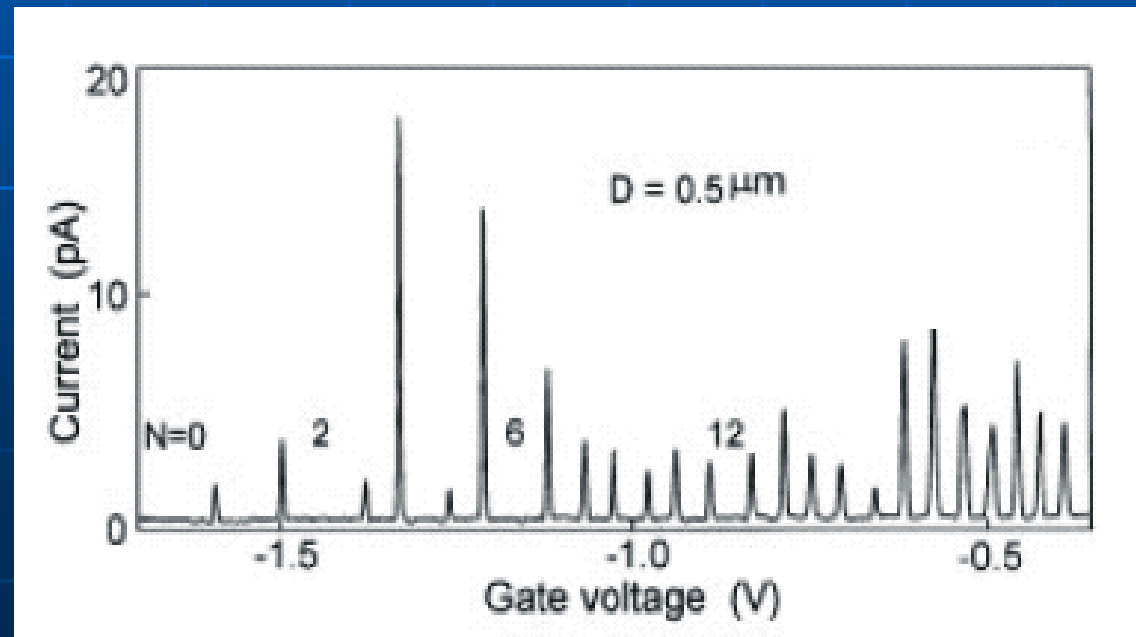
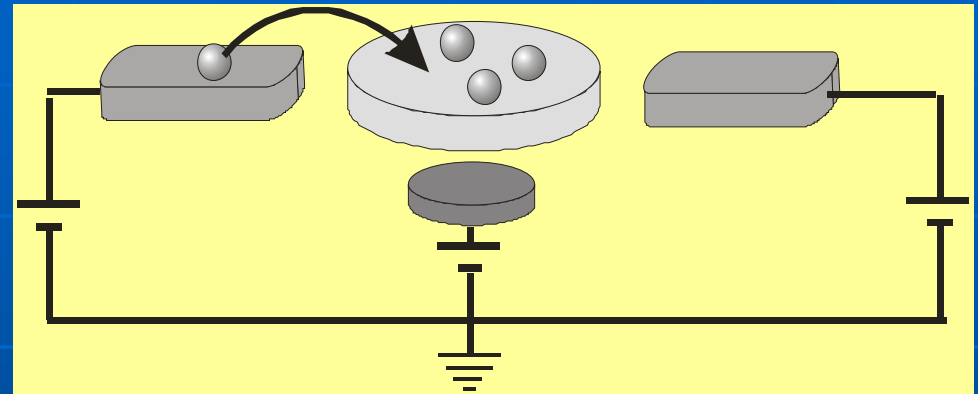
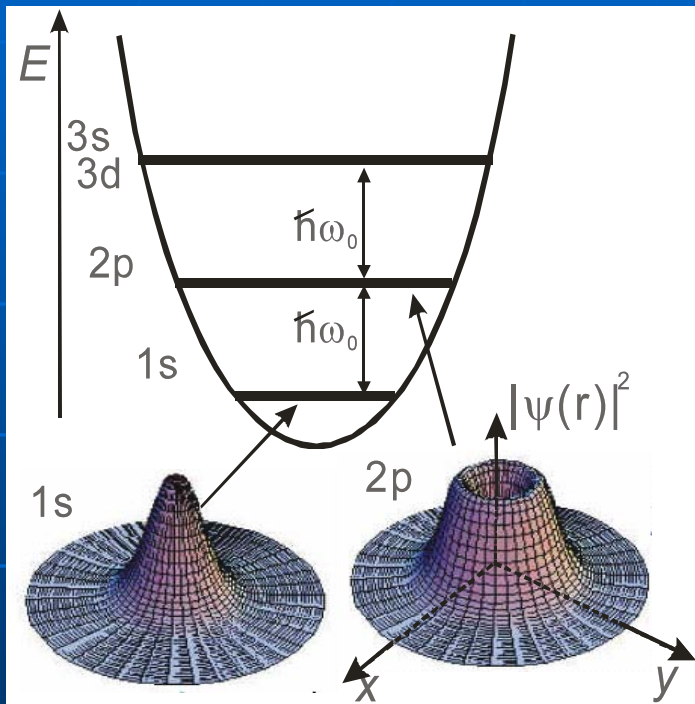
The electrostatic potential of the island increases as one electron tunnels into the island

=> next electron cannot enter the island



Coulomb Blockade

# Quantum Dot (Artificial Atom)

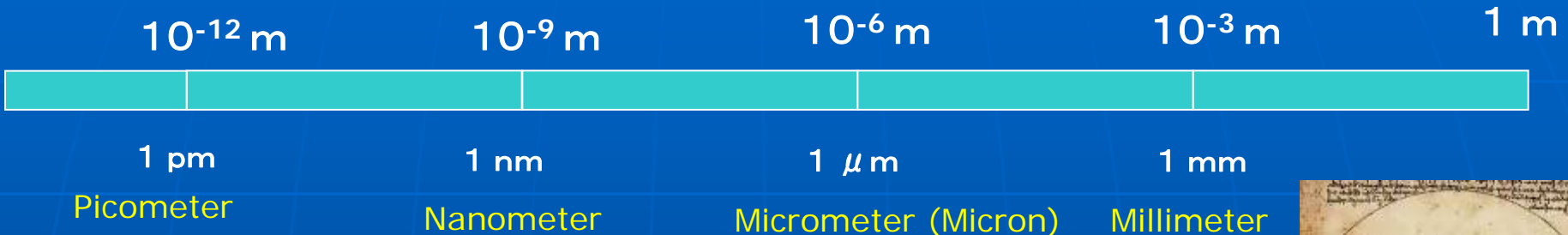


“Periodic table” of artificial atoms

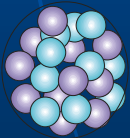
# Observing Atoms

# Manipulating Atoms

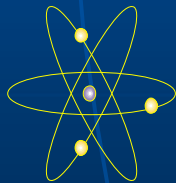
# The Tale of Small World



Size of a nucleus



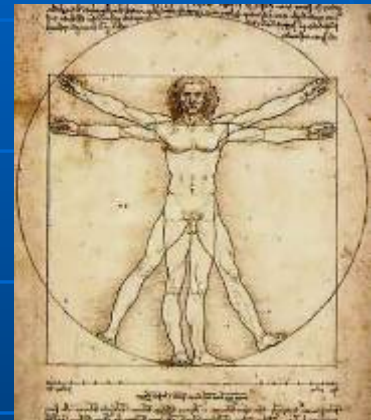
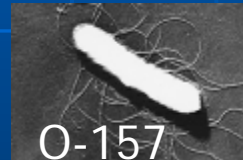
Size of an Atom



Virus



Bacteria



Wavelength of visible light

Optical Microscope



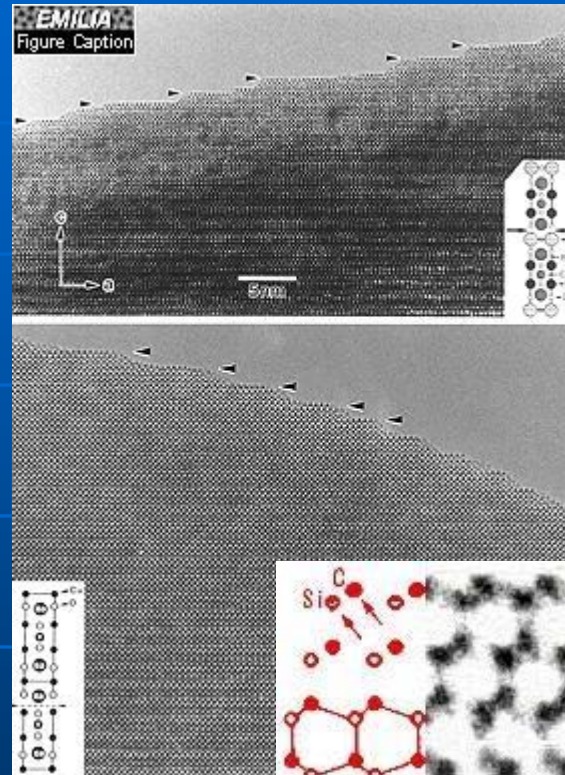
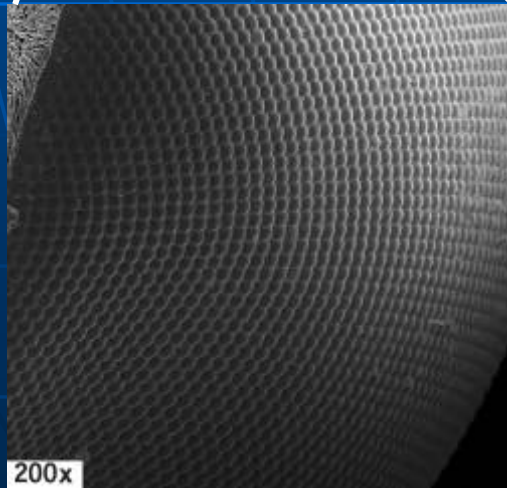
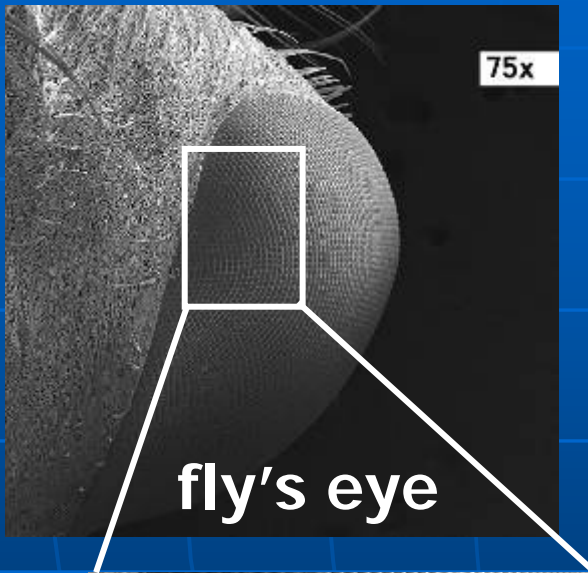
Electron Microscope



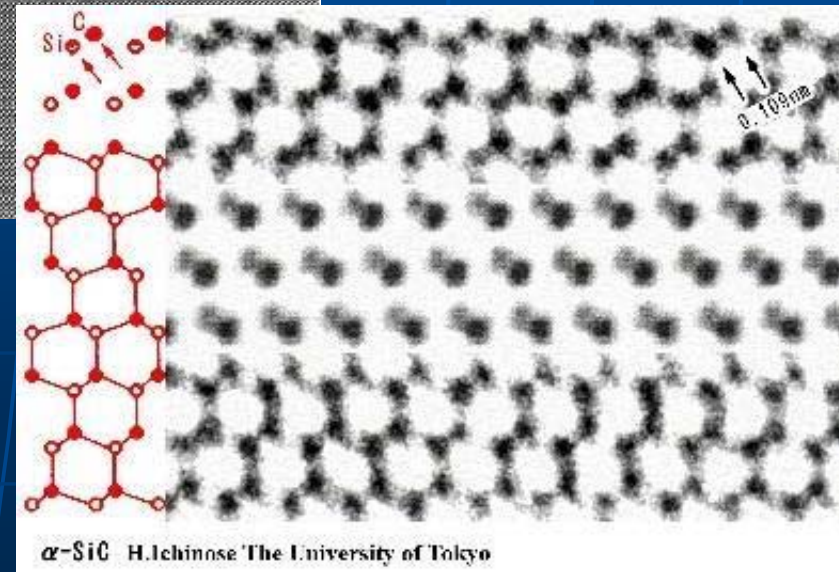
Scanning Probe Microscope



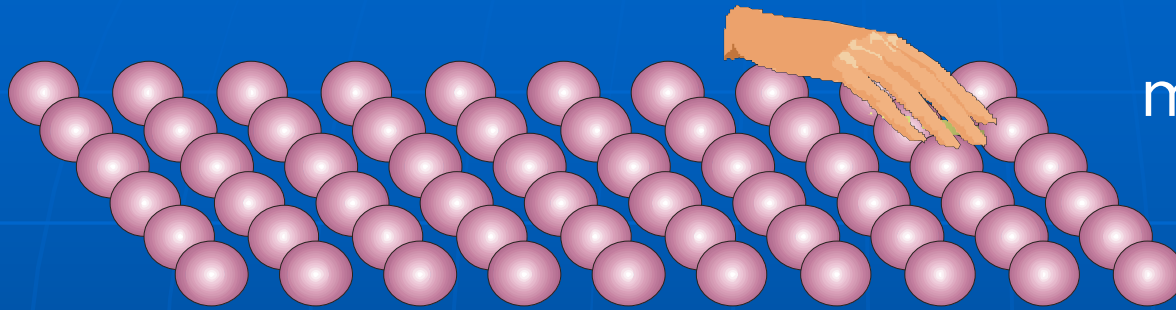
# Seeing Small Objects : Electron Microscope



Observe the regular array of atoms with a high-resolution electron microscope



# Observing the Arrangement of Atoms on a Solid Surface



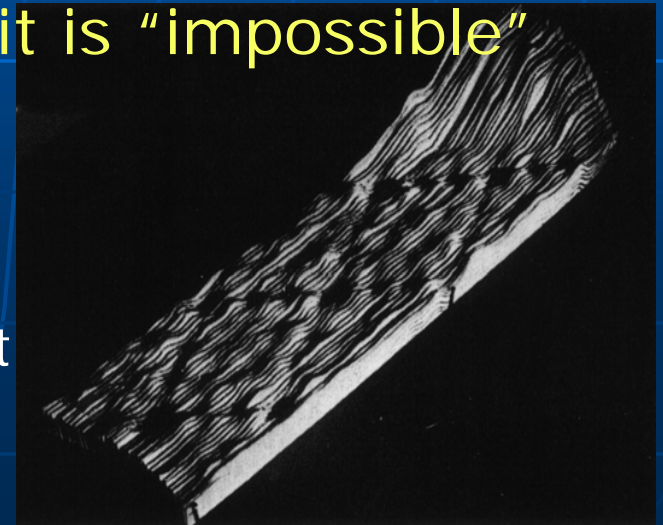
Were it an array of macroscopic objects you can see (or feel) the array by touching it

Would it be possible to achieve the same thing to an array of atoms?

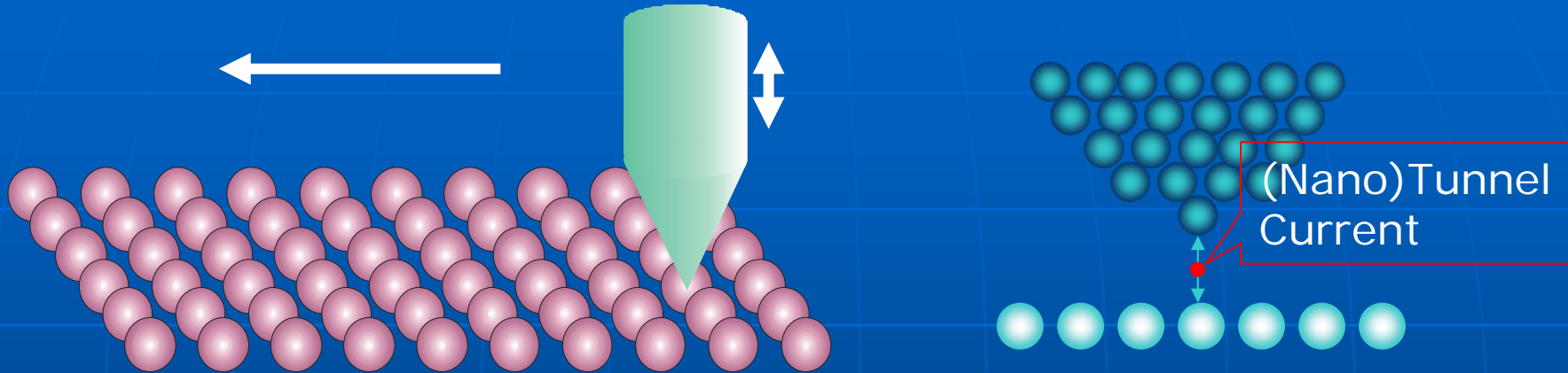
=> common sense tells us that it is "impossible"

Scanning Tunneling Microscope  
1984 Binnig and Rohrer

The first image of atomic arrangement on the surface of a silicon crystal.



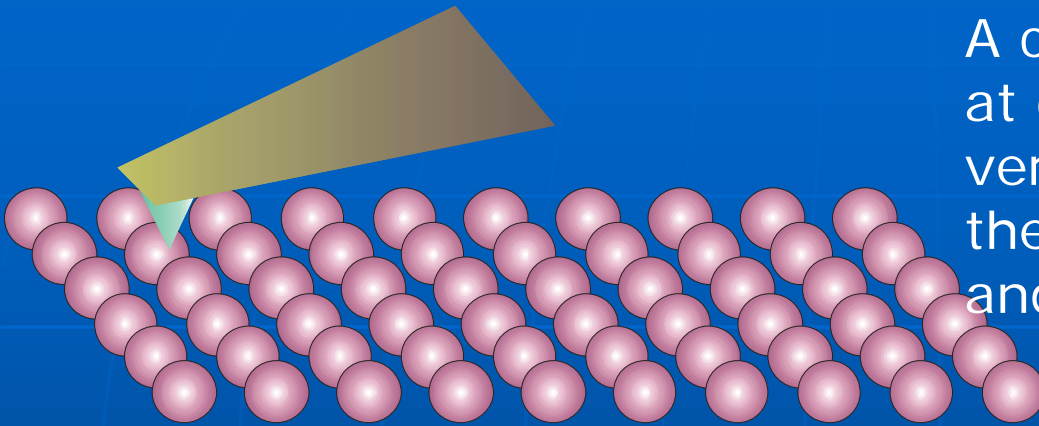
# Scanning Tunneling Microscope (STM)



When atoms at the tip of the needle and atoms on the surface are brought to within  $\sim 1$  nanometer, tunnelling current flows. By moving the tip laterally while keeping the tunneling current constant, one can detect the microscopic undulation due to the array of surface atoms.

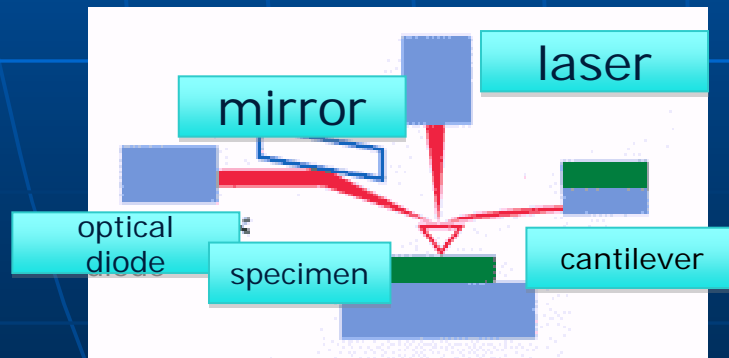
Problem of Stability  
Mechanical Vibration  
Electrical Noise

# Atomic Force Microscope (AFM)

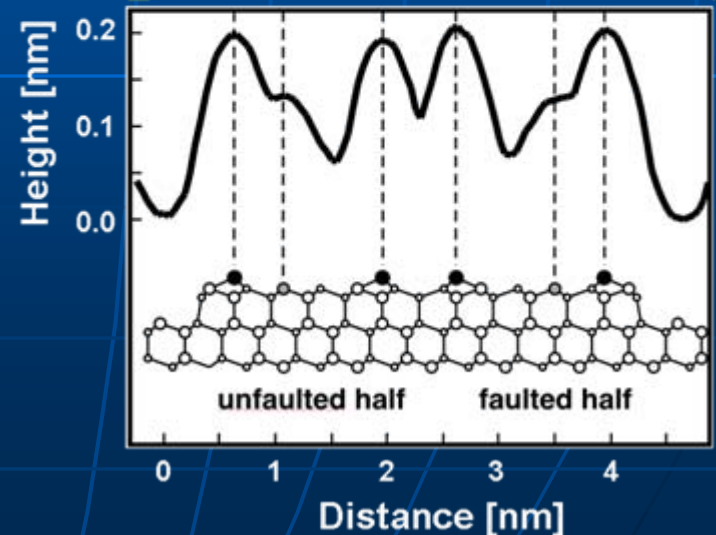
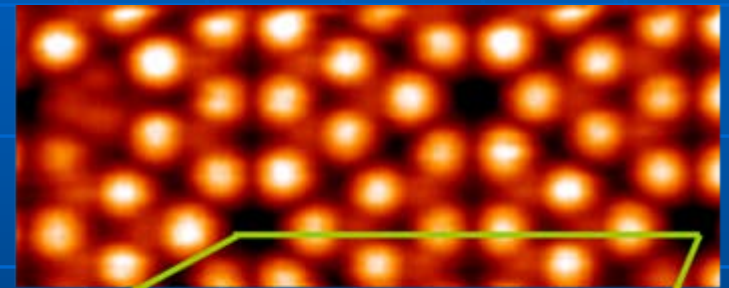


A cantilever (a flexible beam fixed at one end) probe is used to detect very weak forces acting between the atom at the tip of the probe and atoms on the surface.

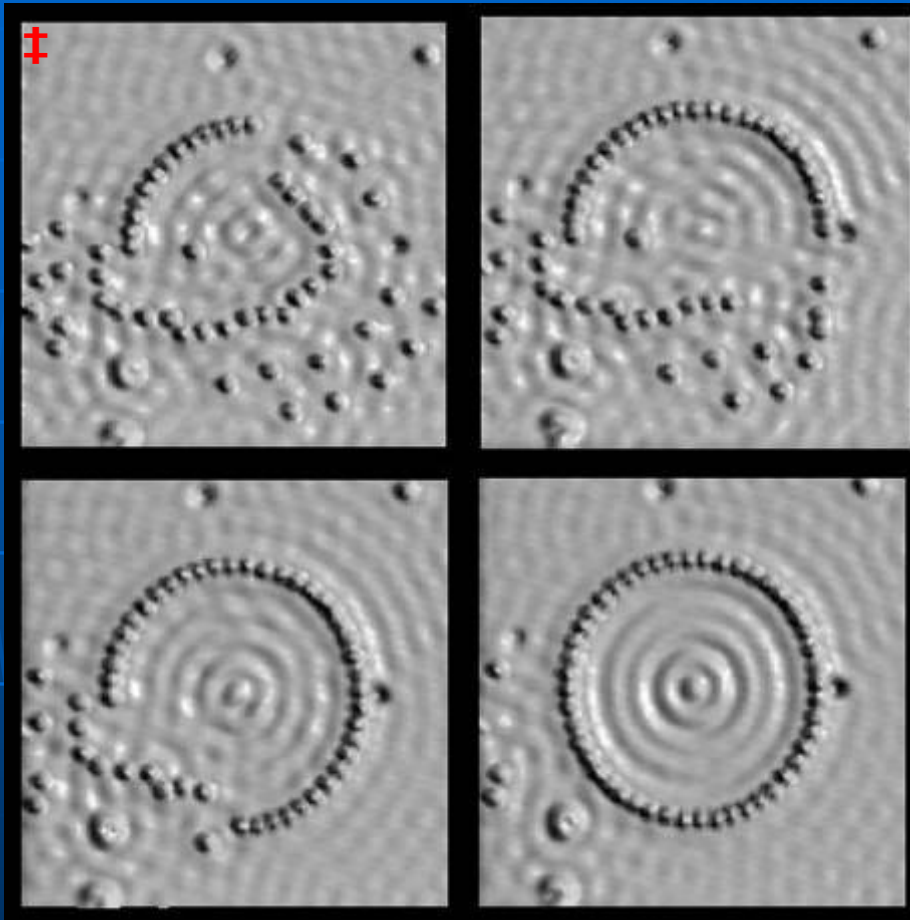
Detect the change in the atomic force as the tip is brought close to and scanned over the surface.



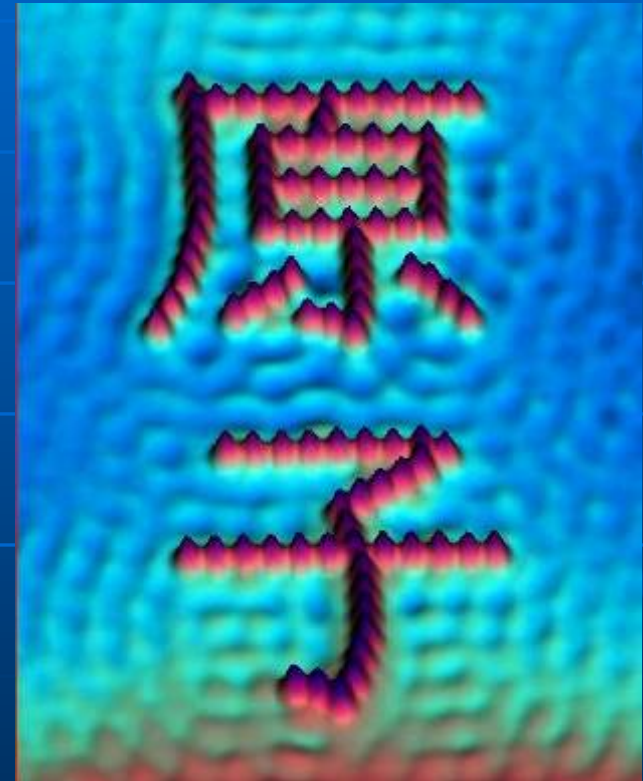
Deflection of the cantilever is detected by the change in the reflected laser beam.



# Manipulating Atoms



IBM Almaden Lab  
Eigler Group



Artificial arrangement iron atoms on a copper surface.  
The ripple-like feature reflects the wave interference  
of the surface electrons

# Macroscopic Quantum Phenomena

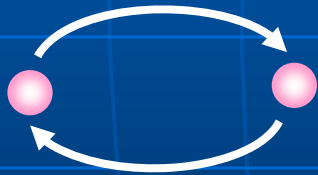
## ---- Bose Condensation and Superconductivity

# Quantum Mechanical Particles

Quantum mechanical particles of the same type are intrinsically indistinguishable.

Even when two particles are interchanged, the state remains the same as before.

The wave function, however, is multiplied by a numerical factor.



$$\Psi(b,a) = C\Psi(a,b)$$

$$\Psi(a,b) = C\Psi(b,a) = C^2\Psi(a,b)$$

$$\Rightarrow C^2 = 1$$

$$\Rightarrow C = 1 \text{ or } -1$$

Boson

Fermion

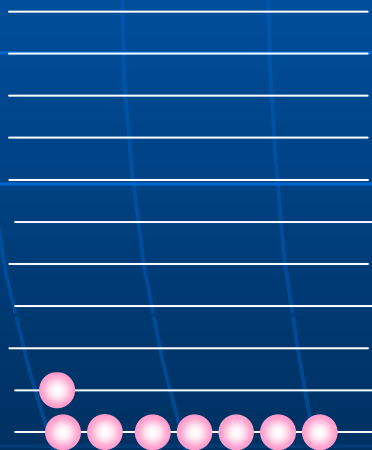
# Quantum Statistics

## Bose Particle (Boson)

Spin: 0, 1, ...



$$\Psi(b,a) = \Psi(a,b)$$



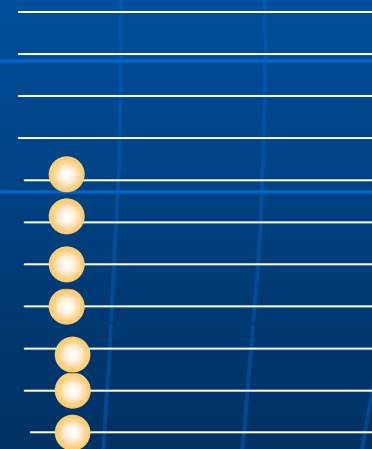
Any number of particles  
can occupy the same state

## Fermi Particle (Fermion)

Spin:  $\frac{1}{2}$ ,  $\frac{3}{2}$ , ...



$$\Psi(b,a) = -\Psi(a,b)$$



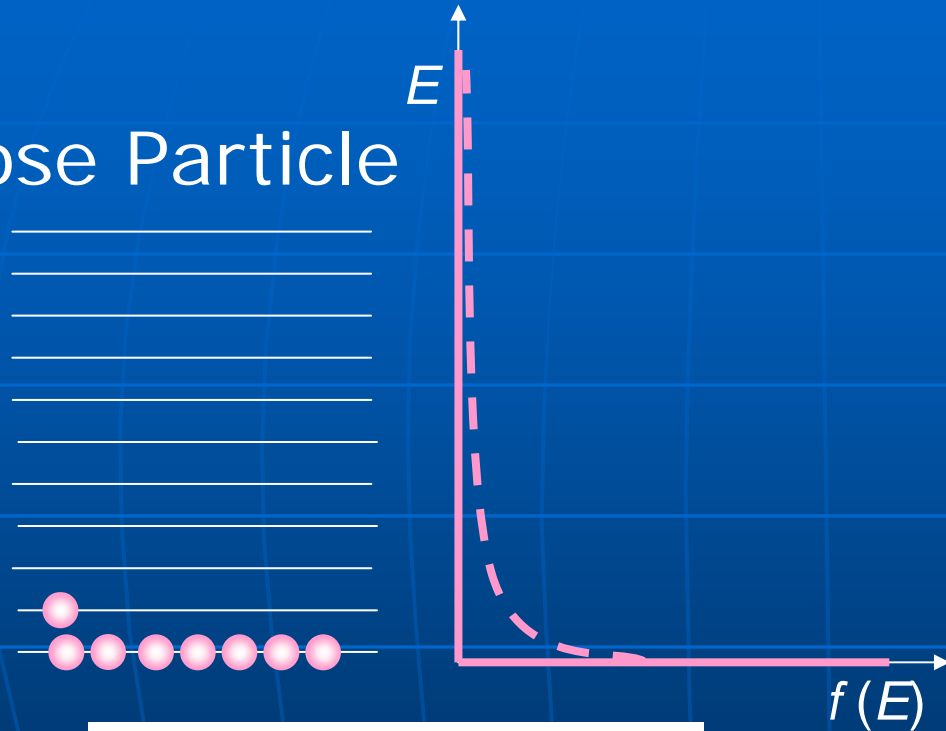
Only one particle is allowed in  
each state (Pauli Exclusion Principle)

If  $a=b$  then

$$\Psi(a,a) = -\Psi(a,a)$$
$$\Rightarrow \Psi(a,a) = 0$$

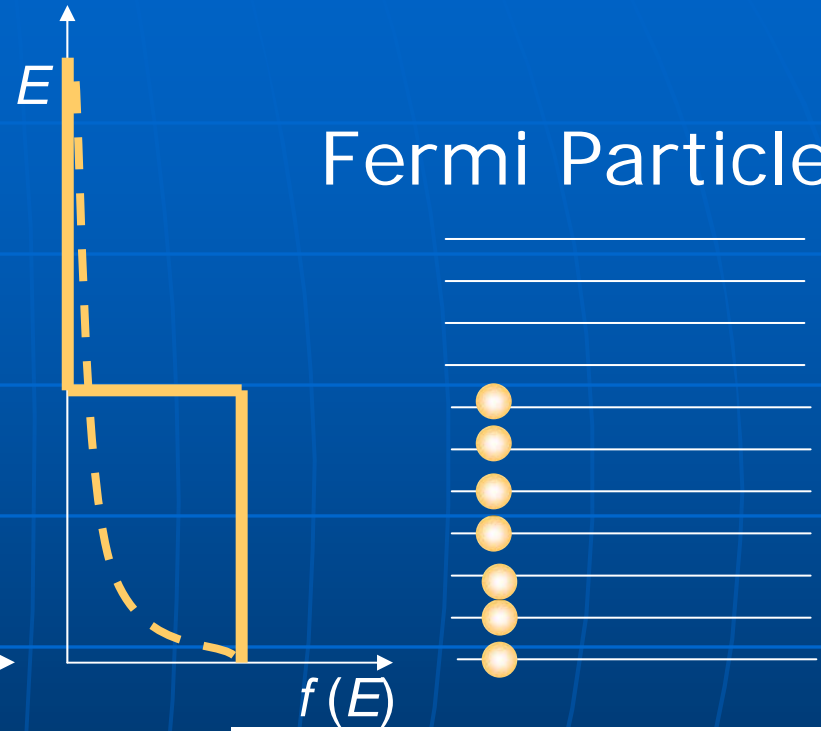
# Bose-Einstein Distribution and Fermi-Dirac Distribution

Bose Particle



$$f_{\text{BE}}(E) = \frac{1}{e^{(E-\mu)/k_{\text{B}}T} - 1}$$

Fermi Particle

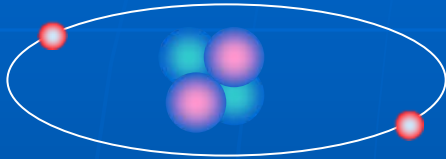


$$f_{\text{FD}}(E) = \frac{1}{e^{(E-\mu)/k_{\text{B}}T} + 1}$$

Either distribution is reduced to the classical Maxwell-Boltzmann distribution at high enough temperatures

$$f(E) = e^{-(E-\mu)/k_{\text{B}}T}$$

# Helium Isotopes

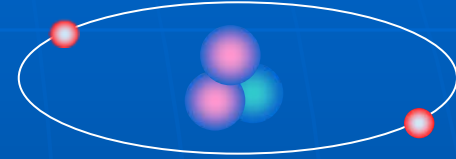


${}^4\text{He}$

2 Protons  
2 Neutrons  
2 Electrons

Total Spin = 0

**Bose Particle**



${}^3\text{He}$

2 Protons  
1 Neutrons  
2 Electrons

Total Spin = 1/2

**Fermi Particle**

# Making Ultra-Low Temperatures

Ultra-low temperatures are required to observe phenomena associated with quantum statistics



Cut-away view of a liquid helium vessel

Liquid Nitrogen 77K

Liquid Helium ( $^4\text{He}$ ) 4.2K

forced evaporation by pumping

$\sim 1.2\text{K}$

Liquid Helium ( $^3\text{He}$ ) 3.2K

forced evaporation by pumping

$\sim 0.3\text{K}$

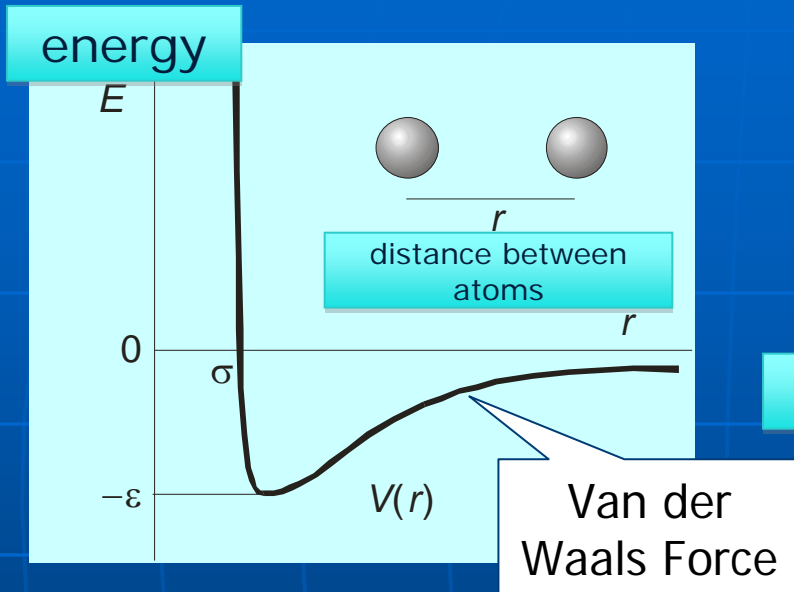
$^3\text{He}$ - $^4\text{He}$  Dilution Refrigerator  $\sim \text{mK}$

Adiabatic Nuclear Demagnetization

$\sim \mu\text{K}$



# Phase Diagram of Helium

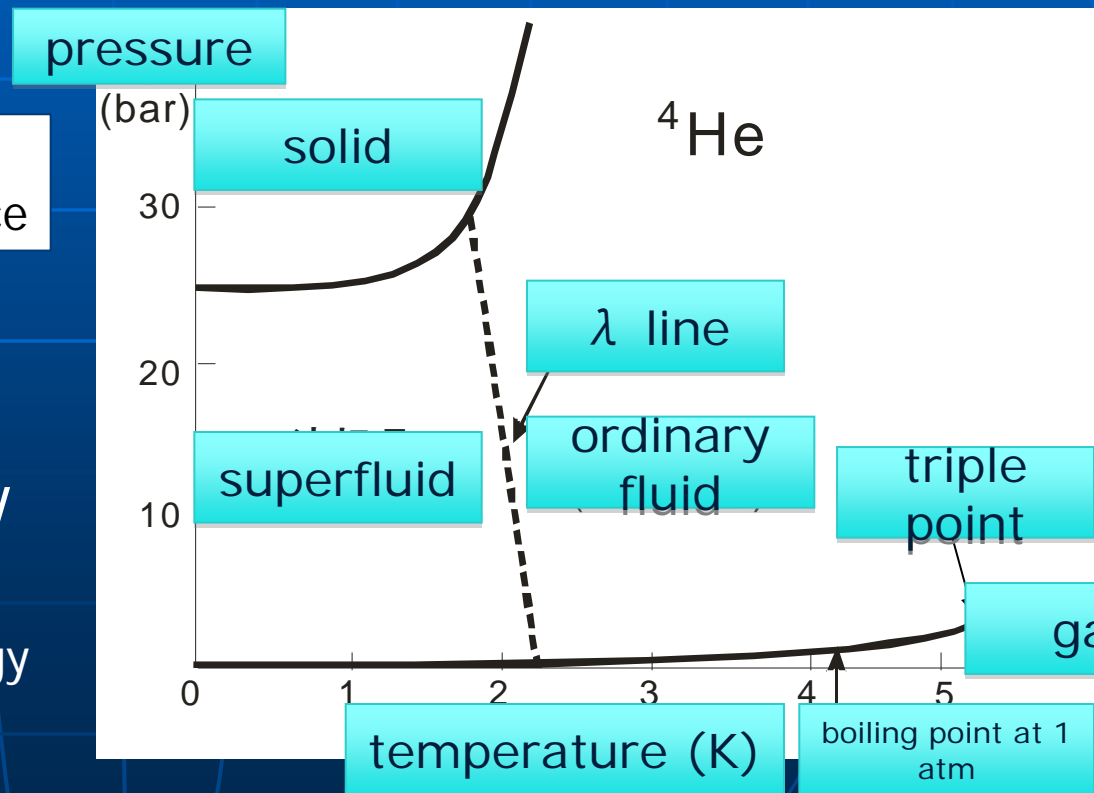


Helium (at ambient pressure) does not solidify even at absolute zero  
=> Quantum Liquid

Helium atoms are light.

Helium atoms interact only weakly with each other.

Kinetic energy > Interaction energy



# Superfluidity of Liquid Helium



The University of Tokyo Cryogenic Research Center †

# Superfluidity of Liquid Helium

figure 1

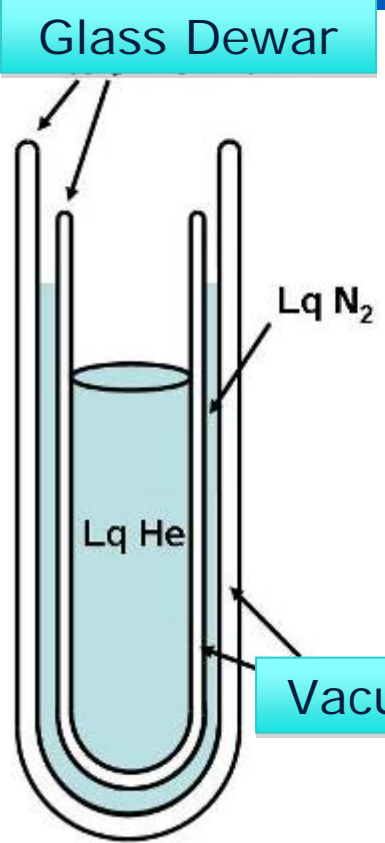
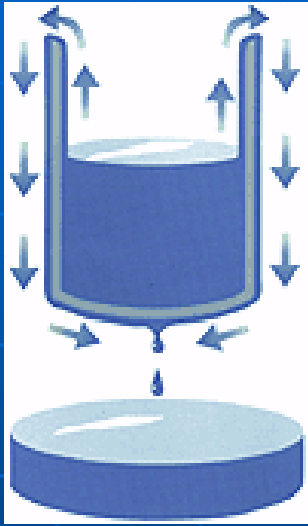


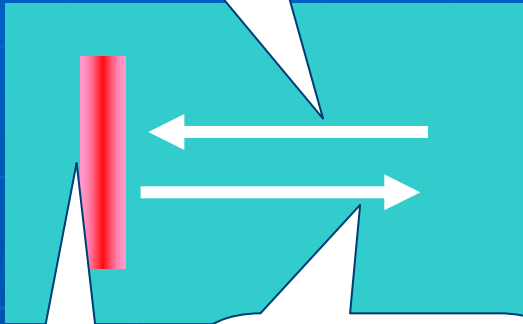
photo 1



# Two-Fluid Model

Thermomechanical Effect (Internal Convection)

Superfluid Component



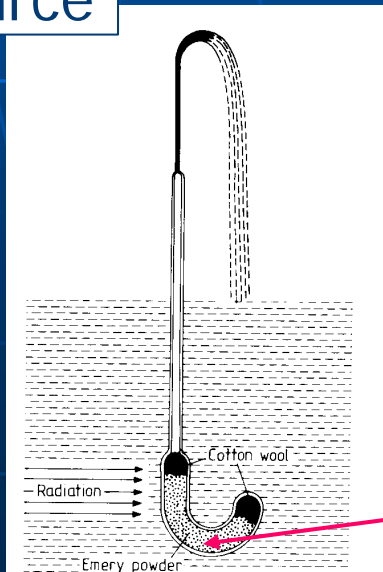
Heat Source

Normal fluid Component



Fountain Effect

the University of Tokyo Cryogenic Research Center



Superfluid component flows without friction through a capillary (superleak) which is too narrow for the normal fluid to pass.

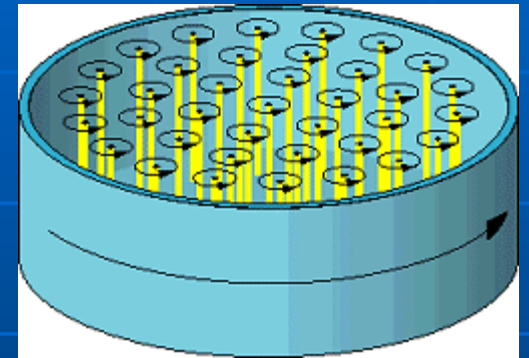
# Quantum Vortex

Macroscopic Wave Function  $\Psi = \Psi_0 e^{i\theta}$

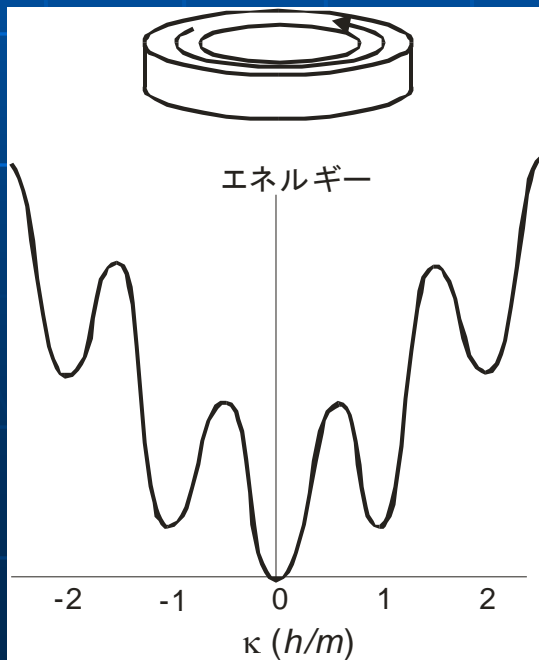
Quantization of Circulation

$$\kappa \equiv \oint_C \mathbf{v}_s \cdot d\mathbf{s} = \frac{\hbar}{m} \oint_C \nabla \theta \cdot d\mathbf{s} = \frac{\hbar}{m} 2\pi n = n \frac{h}{m}$$

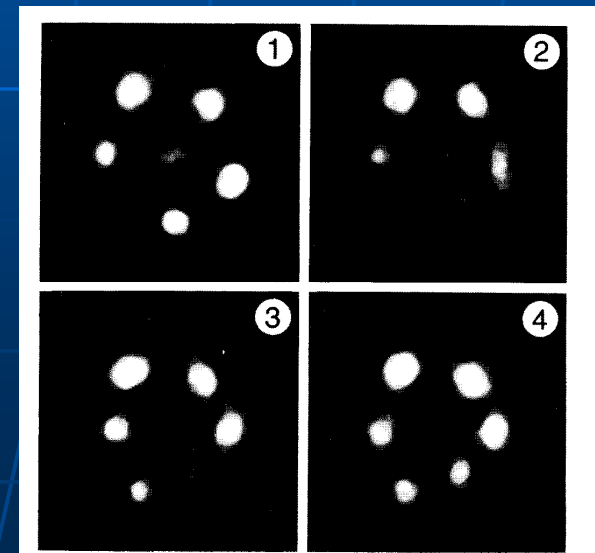
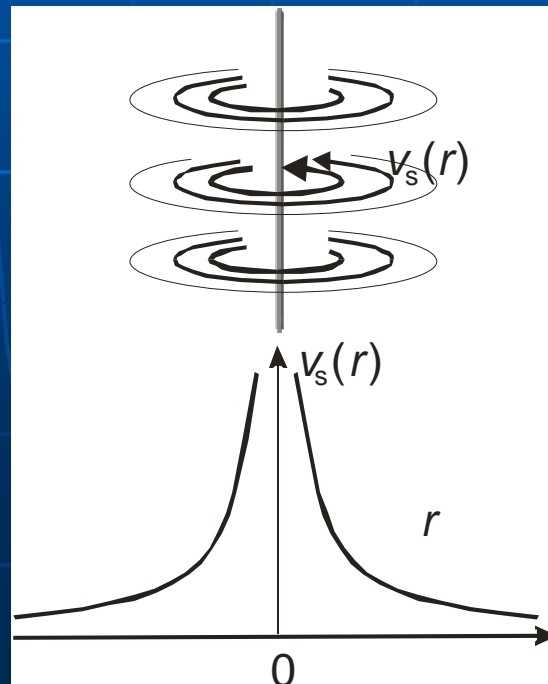
Rotating Bucket Experiment



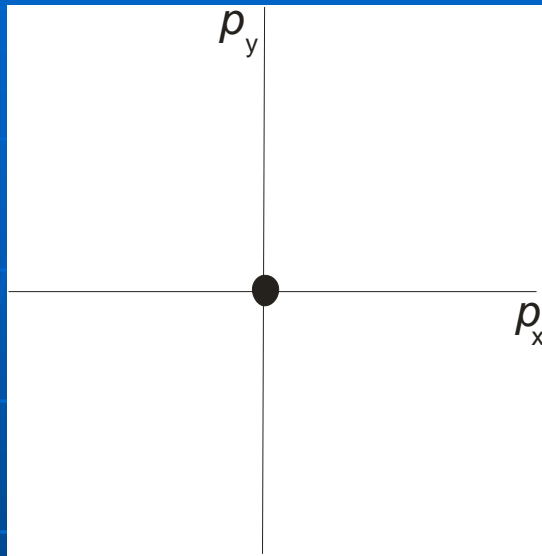
Permanent Current



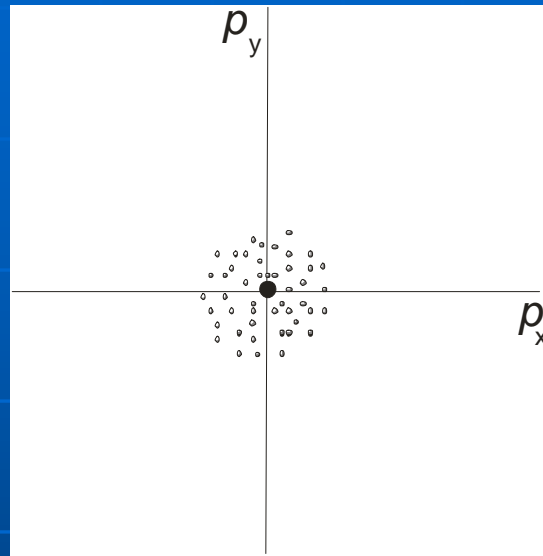
Quantum Vortex



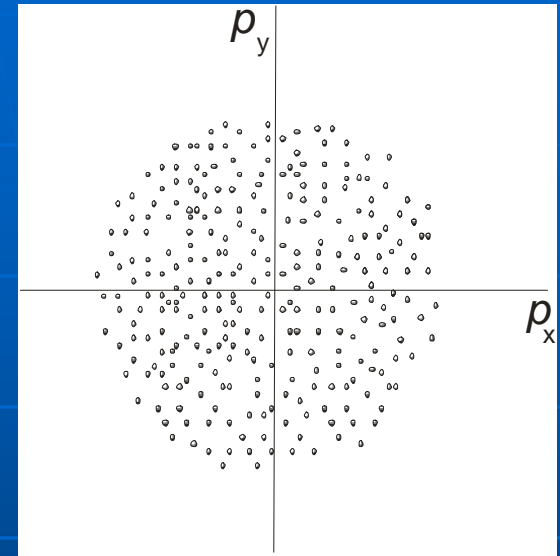
# Bose-Einstein Condensation



$$T = 0$$



$$T = T_{BE}$$



$$T > T_{BE}$$

Thermal de Broglie Wavelength

$$\lambda_T = \left( \frac{2\pi \hbar^2}{mk_B T} \right)^{1/2}$$

Bose condensation occurs when the thermal de Broglie wavelength becomes comparable to the interparticle distance

$$\lambda_T \approx n^{-1/3}$$

$$T_{BE} = \frac{2\pi \hbar^2}{mk_B} \left( \frac{n}{2.612} \right)^{2/3}$$

# Laser Cooling of Atomic Gases

Collect and cool an atomic (*e.g.* Rb) gas (vapor) in a trap

## Doppler Cooling



- Use a laser light with frequency slightly below the resonance frequency of the atoms
- For an atom travelling in the opposite direction to the light, the Doppler shifted light frequency becomes closer to the resonance frequency, so the absorption probability is higher. The momentum of the atom is reduced by absorbing the incident light (photon).
- When the atom reemits the light, it does so isotropically, so that on average the atom is slowed down.
- By a set of six laser beams along the x-, y-, z- axes, Doppler cooling is achieved for all directions.
- With this technique, temperature on the order of  $\sim 100\mu\text{K}$  is attained.  
=> Further reduction of temperature by 3 to 4 orders of magnitude is needed.

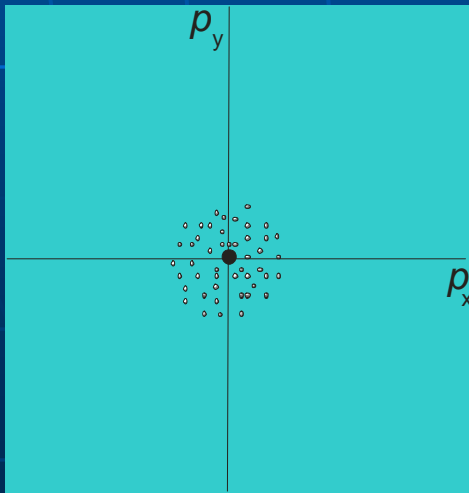
# Bose-Einstein Condensation of Atomic Gases

Reducing the temperature and achieving the conditions of Bose-Einstein Condensation by **Evaporative Cooling** of atomic gas cooled in a magneto-optical trap  
 $T \sim 10^{-7}\text{K}$

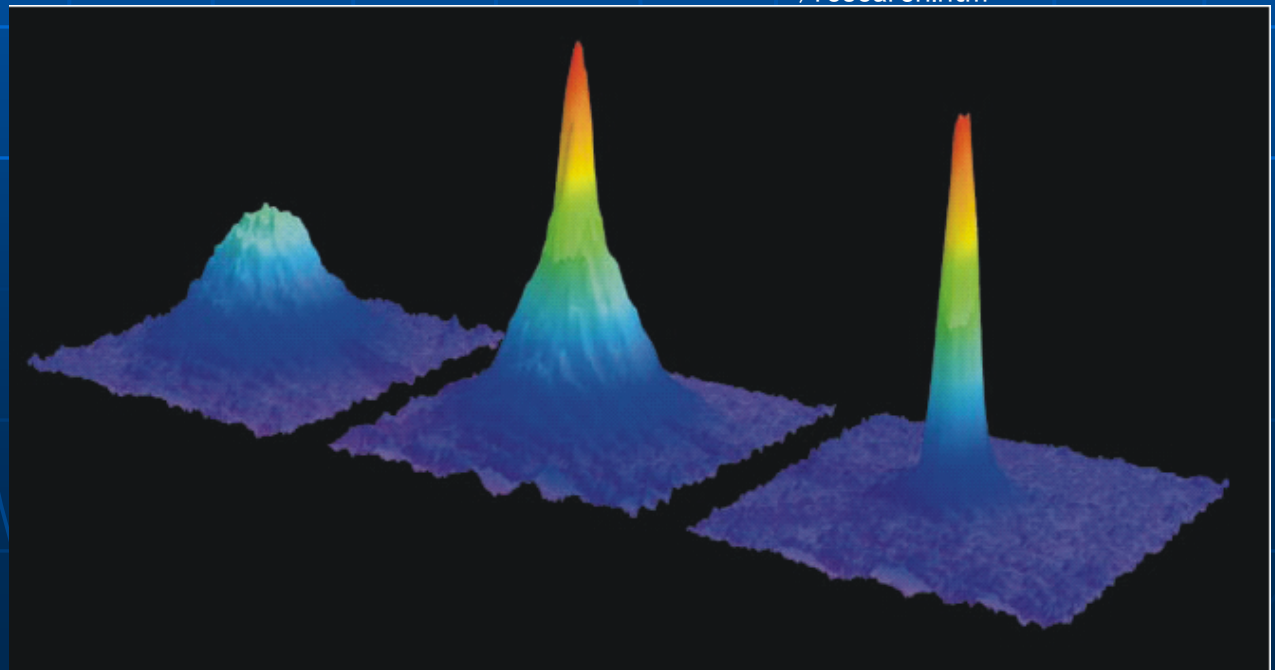
$$\lambda_T \approx n^{-1/3}$$

Upon turning off the trap potential, the cloud of cold atoms drops with gravity, and at the same time swells according to the velocity distribution.

 [http://cua.mit.edu/ketterle\\_group/research.htm](http://cua.mit.edu/ketterle_group/research.htm)

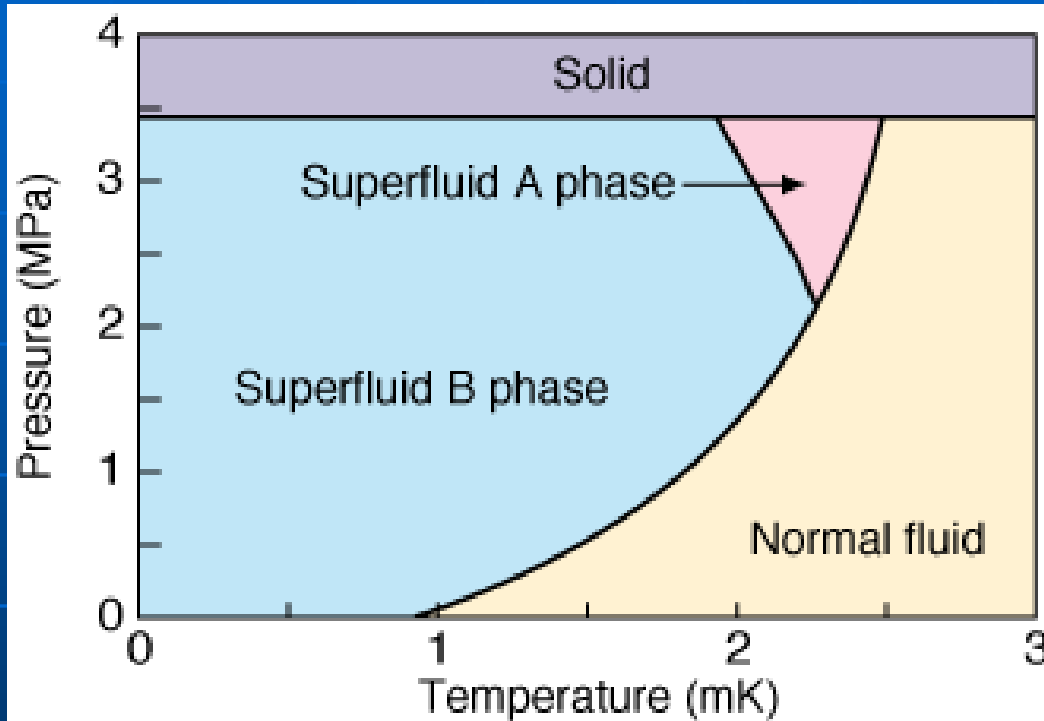


$$T = T_{\text{BE}}$$



# Superfluidity of Liquid $^3\text{He}$

## Phase diagram of $^3\text{He}$

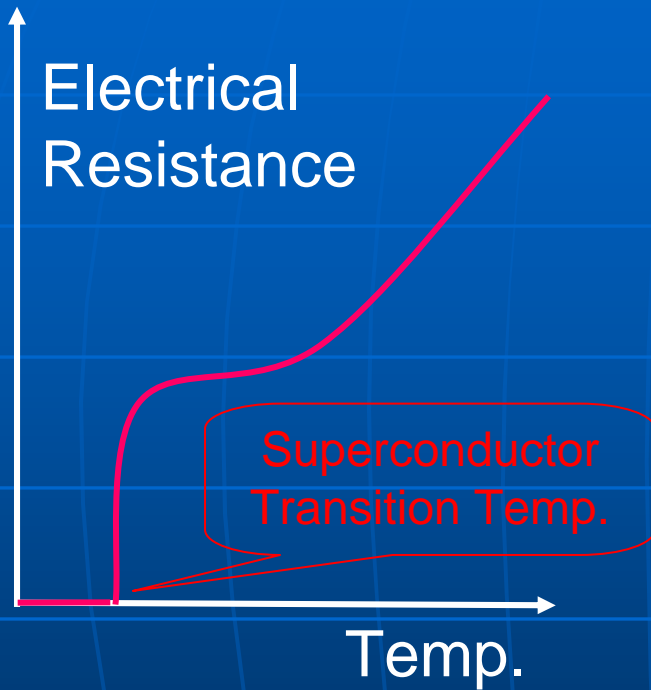


Being Fermi particles,  $^3\text{He}$  atoms do not undergo Bose condensation per se. However, they can become superfluid by forming pairs. (This is the same mechanism as superconductivity)

$^3\text{He}$  becomes a superfluid at an ultra-low temperature of  $\sim 2\text{mK}$

# Basic Properties of Super-Conduction

Perfect Conductor (Zero Resistance)



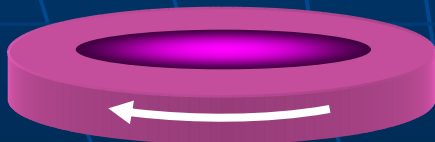
Quantization of Magnetic Flux

$$\phi = n \phi_0$$
$$\phi_0 = \frac{h}{2e} = 2.07 \times 10^{-15} \text{ Wb}$$

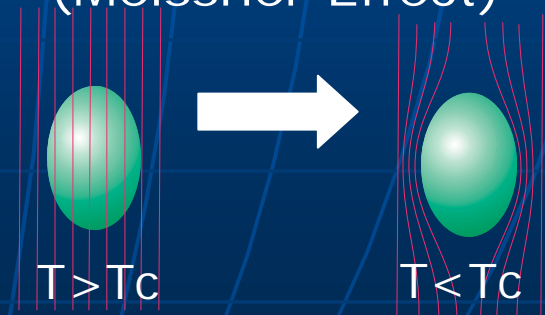
A diagram showing a pink ring with a vertical arrow passing through its center, representing magnetic flux. The equation  $\phi = n \phi_0$  is written to the left of the ring.

Same as quantization of circulation in superfluid

Persistent Electrical Current



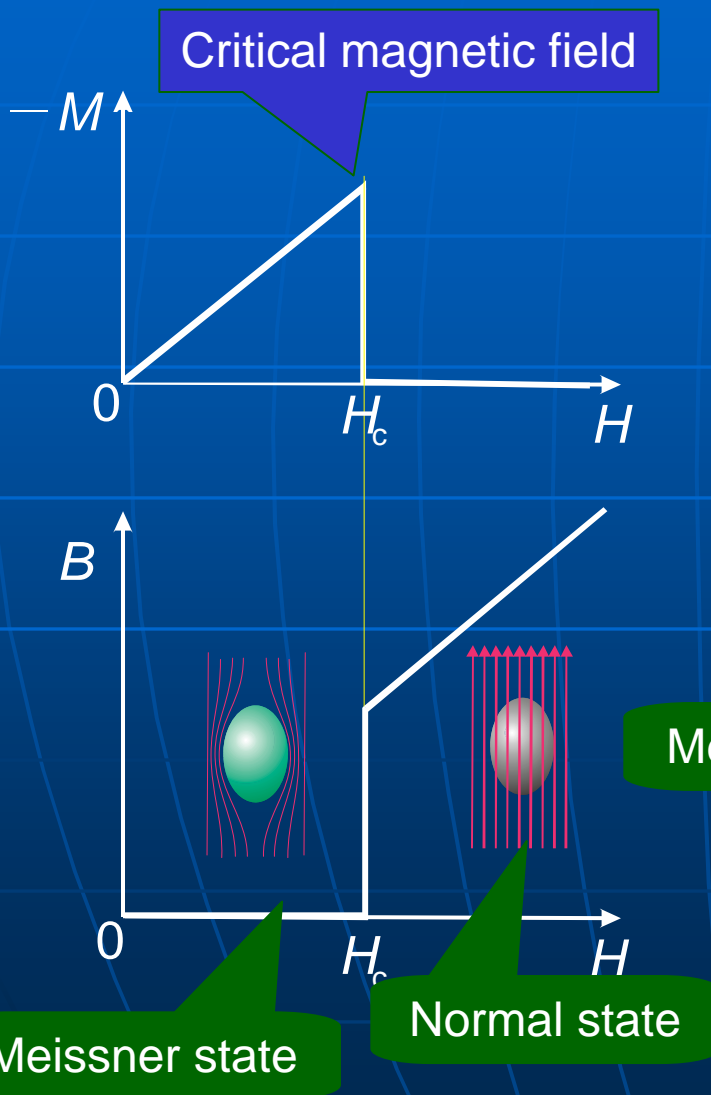
Perfect Diamagnetism (Meissner Effect)



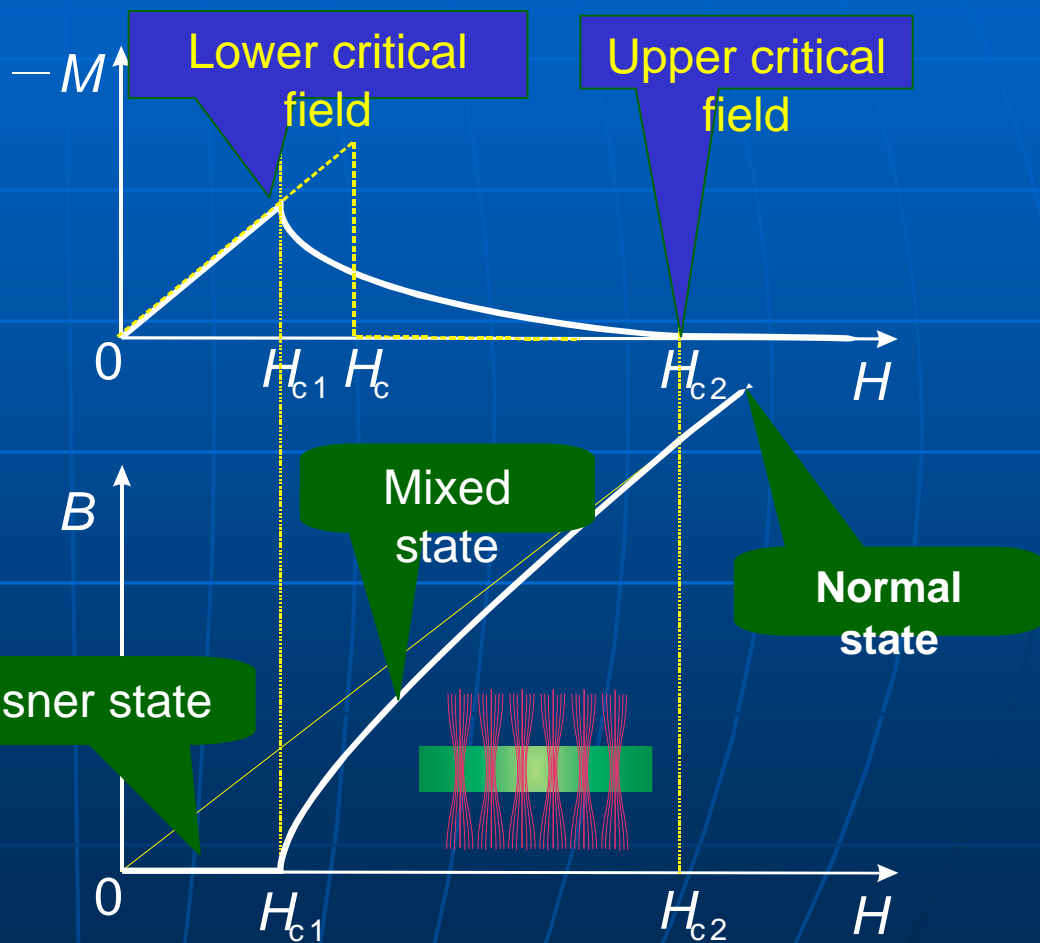
Magnetic flux is expelled from a superconductor

# Type I and Type II Superconductor

## Type I Superconductor



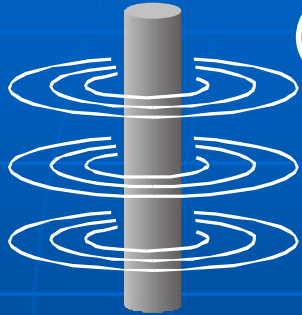
## Type II Superconductor



Superconducting materials of practical use are type II superconductors.

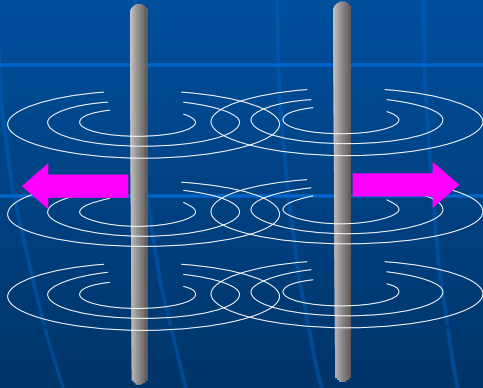
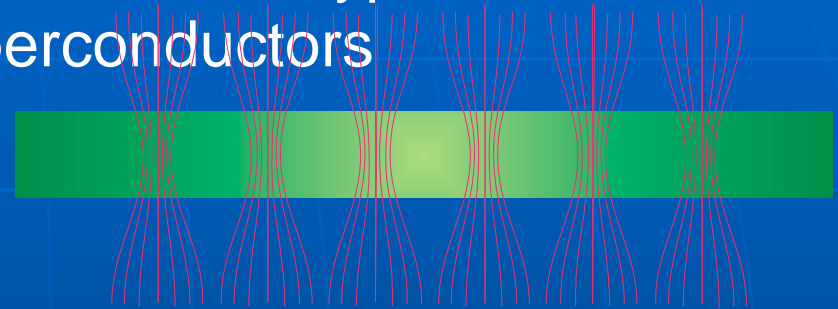
# Quantum Magnetic Flux in Type II Superconductors (Vortex Line)

Quantum magnetic flux  
(Vortex line)



$$\phi_0 = \frac{h}{2e} = 2.07 \times 10^{-15} \text{ Wb}$$

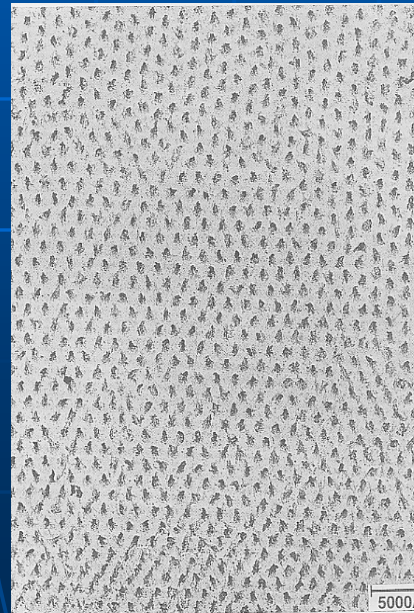
Mixed state of type II  
superconductors



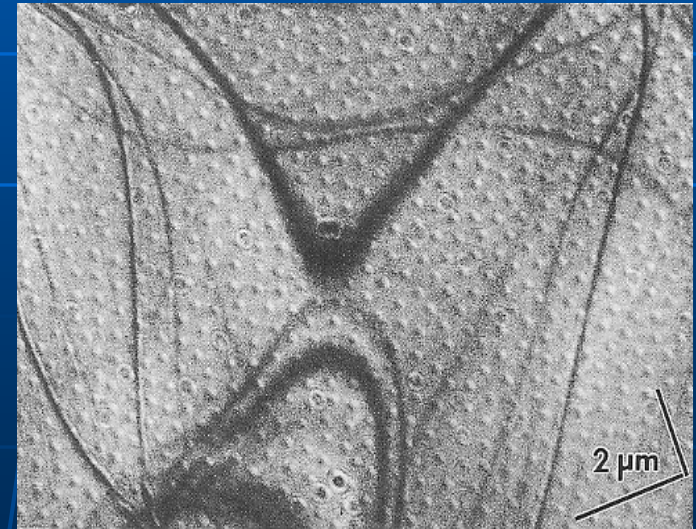
Repulsive force acts between  
vortex lines



Triangular lattice



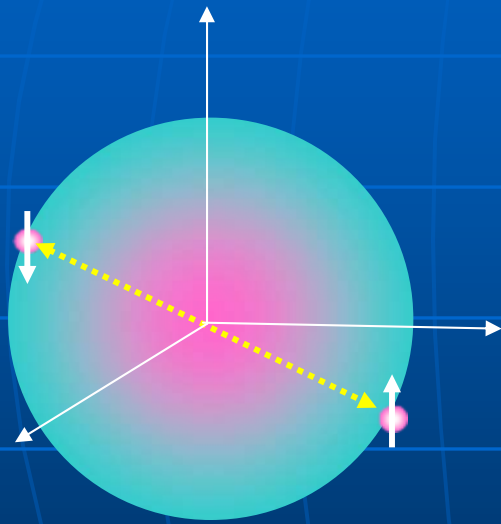
Bitter Method  
(Essmann & Traueble, 1968)



Lorentz Microscope  
(Tonomura Akira, 1992)

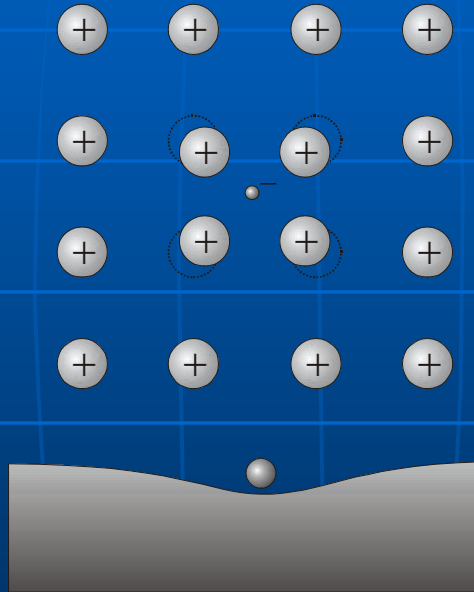
# Mechanisms of Superconductivity

## Cooper Pair Configuration



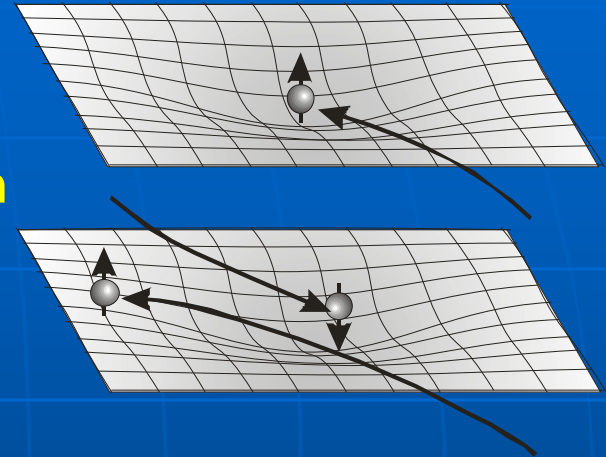
If attractive force acts between two electrons on the Fermi surface, they form a bound state (Cooper Pair).

## Origin of Attraction? Electron Particle Interaction



## Superconducting Transition Temperature

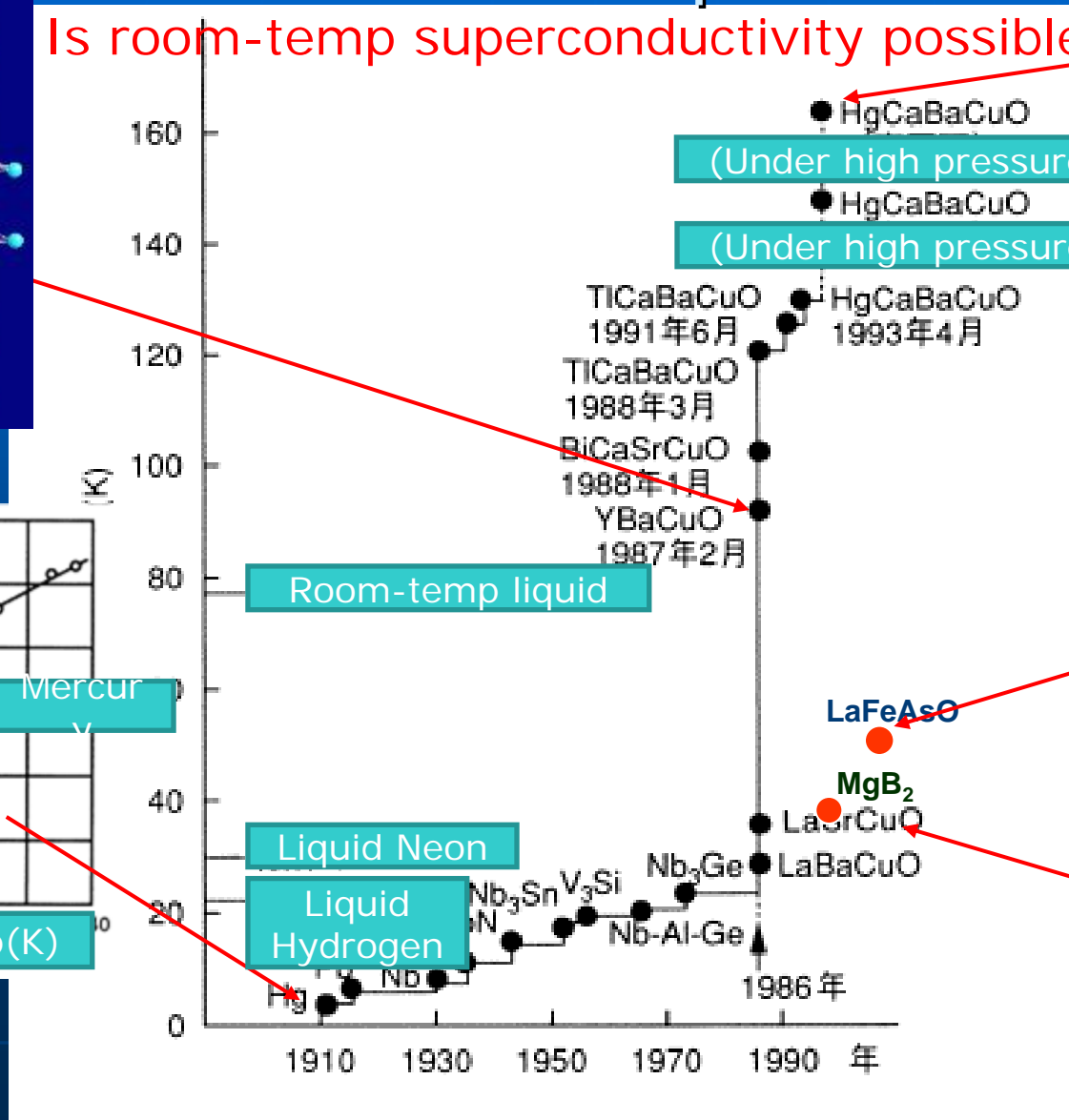
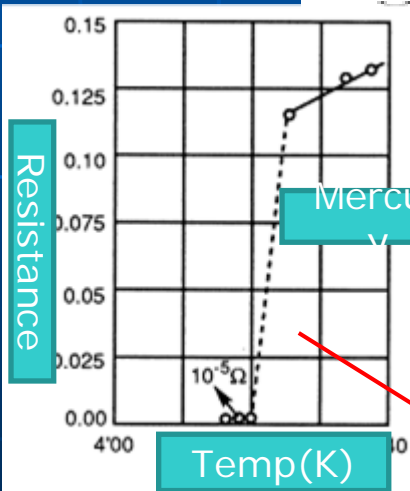
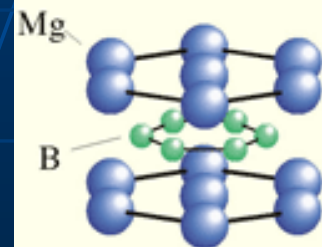
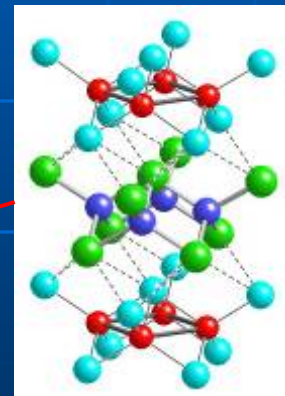
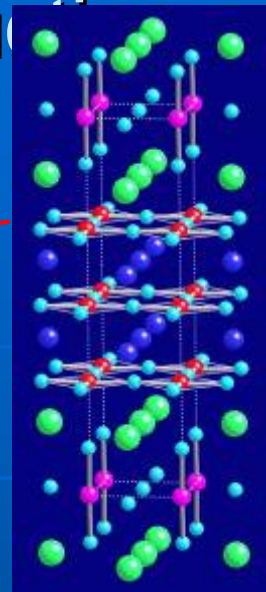
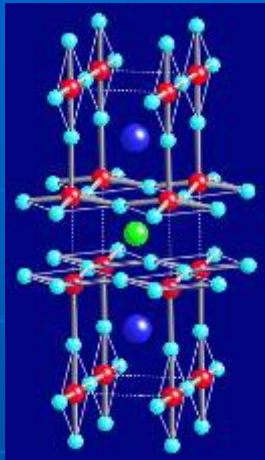
$$T_c = 1.14\Theta_D \exp\left(-\frac{1}{N(0)V}\right)$$



If the inter-electron attractive force mediated by electron-lattice interaction surpasses the repulsive Coulomb force, the net attractive inter-electron interaction results in Cooper pair formation.

# Vicissitude of the Highest Superconducting Transition Temperature

Is room-temp superconductivity possible?





# Summary

- Characteristic of Quantum Mechanics
  - Quantum interference, Tunnel effect
- Mesoscopic Physics
  - Quantum conductance,  $e^2/h$ -physics
  - Quantum Interference AB Effect
  - Single electron tunneling Effect
- Nanotechnology and nanoscience
  - Observing atoms, Manipulating atoms
  - Scanning probe microscope
- Macroscopic Quantum Phenomena
  - Superfluidity
  - Bose Condensation
  - Superconductivity

# The Role of Materials Science and Condensed Matter Physics

- Understanding the diverse properties of various substances based on the basic principles of physics (quantum mechanics and statistical mechanics)
- Phase transition and emergent phenomena:  
seeking universality and unity in diversity and complexity
- Performing experiments that approach the essence of quantum mechanics
- Toward the construction of “materials perspective”: Some of the concepts formed share commonality with other subfields of physics, such as elementary particle physics or astrophysics.
- Forms the basis of engineering

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There are so many interesting things!

Bilateral Pursuit of Humanities and Sciences!